

Synthesis and Anticonvulsant Activity of Oxadiazole Derivatives Incorporating Phthalazin-1,4 (2H, 3H)-dione Scaffolds

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Abstract

In this present work, we prepared the 2- (2-mercapto oxadiazole phthalazin (2H, 3H) -1,4-dione (1) and its salt (2) from 2-hydrazino carbonylmethylphthalazine (2H, 3H) -dione which resulted from condensation of carbondisulfide and potassium hydroxide in ethanol. Also we prepared 2- (2-mercapto oxadiazole-6,7 -dichloro phthalazin (2H, 3H) -1,4-dione (3) and its salt (4). The two salts (2) and (4) reacted with different alkylchloroacetates to produce compounds (5 a-f) and (6 a-f) these compounds tested as anticonvulsant drugs, its revealed anticonvulsant activity compared with phenobarbitone as refrence drug.

Keywords: Oxadiazole • Phthalazindione • Anticonvulsant • Phenobarbitone

Introduction

The phthalazindione nucleus it is promising nucleus, where it used anticancer, anticonvulsant, antibacterial, antiviral, antihyperlipidemic, analgesics, etc. Also the phthalazindione is bioisostere for quinoxalindione and quinazolinindione these compounds either have the dione or one or without i.e., may be phthalazine, quinoxaline and quinazoline [1-23]. These rings formed of benzene ring fused with diazine ring to for phthalazine, quinoxaline and quinazoline which bind with receptor via hydrogen bonds or by bi-bi stacking bond and may be hydrogen bond via carbonyl gp (s) if present in addition to oxadiazole ring which also bind with receptor via heteroatoms which containing it. All three nucleuses phthalazindione, quinoxalindione and quinazolinindione have many and more of biological uses in previous literature [1-7,12-19,23]. All three nucleuses phthalazindione, quinoxalindione and quinazolinindione have many and more of biological uses in previous literature (Figures 1 and 2).

Firstly the starting materials are phthalicanhydrite and 4,5-dichlorophthalicanhydrite which refluxed with hydrazine hydrate in ethanol for six hours to produce phthalazinedione and 6,7 dichlorophthalazinedione which they treated with alcoholic potassium hydroxide gave the salts of phthalazinedione and 6,7-dichlorophthalazinedione they condensed with ethylchloroacetate to obtain on 2- (2-ethylcarbonylmethylphthalazinedione and 2- (2-ethylcarbonylmethyl-6,7-dichlorophthalazinedione. They are reacted with hydrazine hydrate to produce 2-hydrazino carbonylmethylphthalazine (2H, 3H) -dione and 2-hydrazino carbonylmethyl-6,7- dichlorophthalazine (2H, 3H) -dione [8-11,16,20-22].

Synthesis of 2- (2-mercapto oxadiazole methyl phthalaz-

zine-1,4 (2H, 3H) dione (1)

2- (2-hydrazino carbonyl methylphthalazinedione (0.01 mol) (2.34 g) refluxed in ethanol with carbon isulphide (0.01 mol) (0.76 ml) or excess and potassium hydroxide (0.01 mol) (0.56 g) for 12 hrs. then pour on (HCl – H₂O) (3:1) gave the compound (1) m.p. >300, yield 70% (1.95 g) (Figure 3 and Table 1).

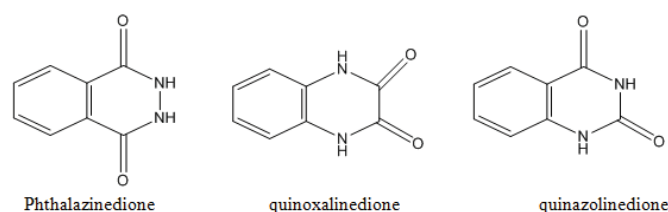


Figure 1. Schematic representation of Phthalazinedione, Quinoxalinedione and Quinazolinindione.

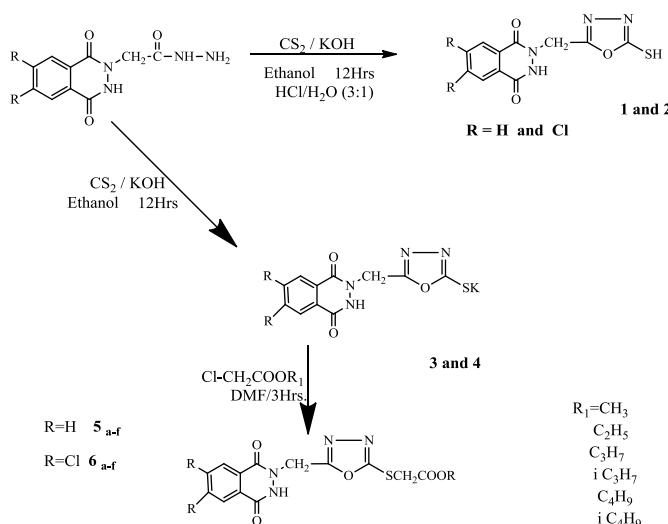


Figure 2. Schematic representation of all three nucleuses phthalazindione, quinoxalindione and quinazolinindione and their biological reaction.

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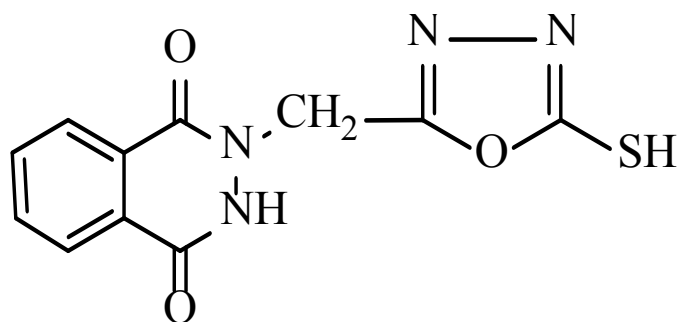


Figure 3. 2-(2-hydrazino carbonyl methylphthalazinedione).

Table 1. Synthesis of 2-(2-mercapto oxadiazole methylphthalazin-1,4-dione) (1).

Variables	C%	H%
Calculated	47.83	2.9
Found	47.36	3.16

Synthesis of 2-(2-mercapto oxadiazole methyl phthalazine-1,4 (2H, 3H)dione potassium salt (2)

2-(2-hydrazinocarbonylmethylphthalazinedione) (0.01 mol) (3.34 g) refluxed in ethanol with carbon-disulphide (0.01 mol) (0.76 ml) or excess and potassium hydroxide (0.01 mol) (0.56 g) for 12 hrs. leave the reaction mixture to cool m.p. >300, yield quantitatively (Figure 4).

Synthesis of 2-(2-mercapto oxadiazole methyl-6,7-dichlorophthalazine (2H, 3H)-1,4-dione (3)

2-(2-hydrazino carbonyl methylphthalazinedione) (0.01 mol) (3.03 g) refluxed in ethanol with carbon disulfide (0.01 mol) (0.76 ml) or excess and potassium hydroxide (0.01 mol) (0.56 g) for 12 hrs. then pour on (HCl - H₂O) (3:1) gave the compound (3), m.p. 288-289, yield 65% (2.26 g) (Figure 5 and Table 2).

Synthesis of 2-(2-mercapto oxadiazole methyl-6,7-dichlorophthalazine (2H, 3H)-1,4-dione potassium salt (4)

2-(2-hydrazinocarbonylmethyl-6,7- phthalazinedione) (0.01 mol) (3.45 g) refluxed in ethanol with carbon disulfide (0.01 mol) (0.76 ml) or excess and potassium hydroxide (0.01 mol) (0.56 g) for 12 hrs. leave the reaction mixture to cool, m.p. >300, yield quantitatively (Figure 6).

2 (5-Alkoxy carbonylmethylthio-1,3,4-oxadiazol-2-yl-methyl) -1,4- (2H, 3H) phthalazinedione (5a-f) and 2 (5-Alkoxy carbonylmethylthio-1,3,4-oxadiazol-2-yl-methyl)-6,7-dichloro-1,4- (2H, 3H) phthalazinedione (6a-f)

A mixture of equimolar quantities 3.15 g, (0.01 mol) of 3 or 3.84 g, (0.01 mol) of 4 and the appropriate alkyl choroacetates in dimethylformamide (50 ml) was heated for two hours on water bath (Figure 7). The reaction mixture was then cooled, poured onto a stirred ice cold water (200 ml) for 30 minutes and the resulting solid was purified, washed with water, dried and recrystallized from ethanol (Tables 3 and 4).

Materials and Methods

Rats of either sex weighing 20-30 g were arranged in groups each of six animals. Phenobarbitone (Aldrich) was used as a reference drug and pentylene-tetrazole (Aldrich) was used to induce convulsions in the experimental animals. Five compounds of phthalazinedione derivatives were selected for evaluation of anticonvulsant activity and their specifications are presented in Determination of the convulsive dose of pentylene-tetrazole. Four graded doses of pentylene-tetrazole (80, 160, 240 and 320 mg/kg) were injected to groups of six frogs. The animals were observed for 60 minutes. The

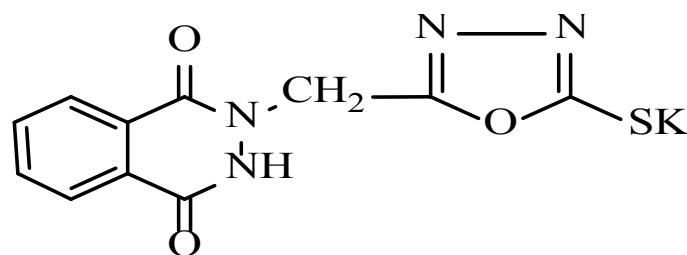


Figure 4. 2-(2-hydrazinocarbonylmethylphthalazinedione).

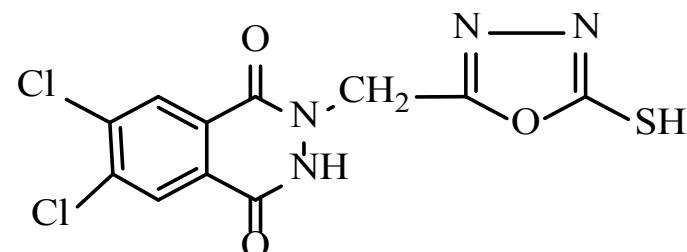


Figure 5. 2-(2-hydrazino carbonyl methylphthalazinedione).

Table 2. Synthesis of 2-(2-mercapto oxadiazole methyl-6,7- dichloro phthalazine (2H, 3H) -1,4-dione (3).

Variables	C%	H%
Calculated	38.26	1.74
Found	38.05	2.10

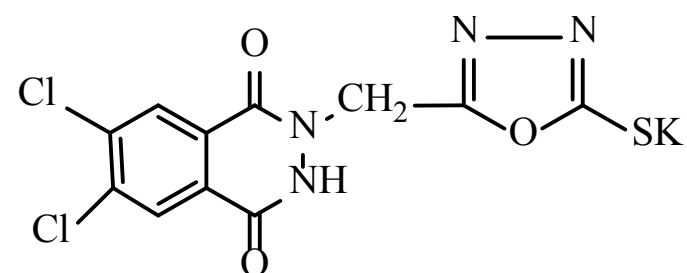


Figure 6. 2-(2-hydrazinocarbonylmethyl-6,7- phthalazinedione).

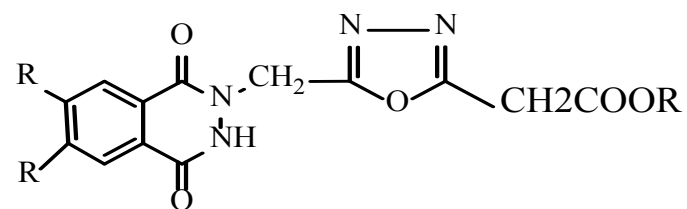


Figure 7. 2-(5-Alkoxy carbonylmethylthio-1,3,4-oxadiazol-2-yl-methyl).

dose of 320 mg/kg was found to be the suitable convulsive dose that could induce a tonic convulsion in all injected animals within 30 minutes and without death during 24 hours.

Results and Discussion

Pharmacological testing

A variety of the newly synthesized phthalazinedione were subjected to preliminary pharmacological testing with regard to their sedative hypnotic as well as anticonvulsant activities. CNS-depressant activity of many sedative-hypnotic agents are mainly evaluated using several screening methods including, loss of righting reflex, corneal reflex, motor activity, amphetamine antagonism and response to the various chemical stimulants. On the other hand most of the experimental methods for evaluation of the anticonvulsant activity of many drugs involve the artificial induction of convulsions by chemo

Table 3: Spectral Data of Some of the Compounds and their elemental analysis.

Comp. No.	R	R ₁	M.P. °C	Yield %	Molecular formula M. Wt	Elemental analyses		
						%	Calcd.	Found
5 _a	H	CH ₃	160-1	68	C ₁₄ H ₁₂ N ₄ O ₅ S 348	C	48.28	47.95
						H	3.45	3.67
						N	16.09	16.03
5 _b	H	C ₂ H ₅	80-2	88	C ₁₆ H ₁₄ N ₄ O ₅ S 362	C	49.72	49.55
						H	3.87	4.10
						N	15.47	15.13
5 _c	H	C ₃ H ₇	125-7	75	C ₁₈ H ₁₆ N ₄ O ₅ S 376	C	51.06	50.85
						H	4.26	3.90
						N	14.89	14.55
5 _d	H	Iso-C ₃ H ₇	90-2	50	C ₁₆ H ₁₆ N ₄ O ₅ S 376	C	51.06	50.77
						H	4.26	4.06
						N	14.89	15.10
5 _e	H	C ₄ H ₉	160-2	65	C ₁₇ H ₁₆ N ₄ O ₅ S 390	C	52.31	51.95
						H	4.62	4.35
						N	14.36	14.11
5 _f	H	Iso-C ₄ H ₉	149-1	75	C ₁₇ H ₁₆ N ₄ O ₅ S 390	C	52.31	51.95
						H	4.62	4.36
						N	14.36	13.88
6 _a	Cl	CH ₃	176-8	80	C ₁₄ H ₁₀ Cl ₂ N ₄ O ₅ S 417	C	40.29	39.95
						H	2.40	2.36
						N	13.43	13.88
6 _b	Cl	C ₂ H ₅	158-9	90	C ₁₅ H ₁₂ Cl ₂ N ₄ O ₅ S 431	C	41.76	42.04
						H	2.78	2.55
						N	12.99	12.65
6 _c	Cl	C ₃ H ₇	130-2	80	C ₁₆ H ₁₄ Cl ₂ N ₄ O ₅ S 445	C	43.15	42.90
						H	3.15	3.05
						N	12.58	12.30
6 _d	Cl	Iso-C ₃ H ₇	122-4	72	C ₁₆ H ₁₄ Cl ₂ N ₄ O ₅ S 445	C	43.15	43.07
						H	3.15	3.01
						N	12.58	12.90
6 _e	Cl	C ₄ H ₉	180-2	70	C ₁₇ H ₁₆ Cl ₂ N ₄ O ₅ S 459	C	44.44	44.75
						H	3.49	3.33
						N	12.20	12.05
6 _f	Cl	Iso-C ₄ H ₉	170-2	78	C ₁₇ H ₁₆ Cl ₂ N ₄ O ₅ S 459	C	44.44	44.04
						H	3.49	3.09
						N	12.20	12.10

Table 4: Spectral Data of Some of the newly Synthesized Compounds.

Comp. No.	Item	IR cm ⁻¹ , ¹ HNMR, ppm, Ms (m/z) Spectral data
3	IR	3165 (NH of the ring), 3017 (CH aliphatic), 1660,1601 (carbonyls of phthalazindione nucleus)
	¹ HNMR	5.67 (s, 2H, NCH ₂ CO), 7.90 (s,1H, aromatic protons), 8.29 (s,1H, aromatic protons) , 9.71 (s, 1H, NH of the ring), 12.06 (s,1H, SH of the oxadiazole group).
mp5 _a	IR	3168 (NH of the ring), 3023 (CH,aliphatic), 1748 (ester carbonyl), 1655 (carbonyl of the ring).
	¹ HNMR	3.71 (s, 3H, COOCH ₃), 5.00 (s, 2H, NCH ₂), 7.92-8.27 (m, 4H, aromatic protons), 11.97 (s, 1H, NH of the phthalazine ring)
5 _b	IR	3169 (NH of the phthalazine ring , 2987 (CH, aliphatic), 1759 (ester carbonyl), 1658, 1593 (carbonyls of phthalazinedione nucleus)
	¹ HNMR	1.19 (t, 3H, COO-CH ₂ -CH ₃), 3.37 (s, 2H, N-CH ₂), 4.15 (q, 2H, CO ₂ CH ₂ CH ₃), 4.96 (s, 2H, SCH ₂ -CO), 7.94-8.30 (m,4H, aromatic protons), 11.98 (s, 1H, NH of the ring).
6 _b	IR	3414 (my be SH of due to resonance), 3161 (NH of the ring), 2991 (CH, aliphatic), 1749 (ester carbonyl), 1652, 1603 (carbonyls of phthalazine ring)
	¹ HNMR	1.19 (t, 3H, COOCH ₂ -CH ₃), 3.91 (s, 2H, N-CH ₂), 4.17 (q, 2H, COOCH ₂ CH ₃), 4.97 (s,2H, SCH ₂ -CO), 8.17-8.42 (2d,2H, aromatic protons of dichloro derivatives of phthalazinedione), 11.75 (s,1H, NH of the ring).
6 _a	MS	417 (M) 419 (M+2)10%,15% Respectively ,344 (100%) (M – CH ₃ COO) and 358 (70%) these results indicate to presence of dichloro in the compounds where the mass spectroscopy revealed the isotopes.
6 _c	MS	445 (M) 443(M-2)40%,60% and 29 (33%) Respectively ,358 (100%) these results indicate to presence of dichloro in the compounds where the mass spectroscopy revealed the isotops.
6 _e	MS	457 (M-2) 459 (M) 7%,10%, 230 (30%) Respectively , 344 (100%) these results indicate to presence of dichloro in the compounds where the mass spectroscopy revealed the isotops.

and / or electroshock agents and inhibition of such convulsions by the drug under test. In this investigation the loss of righting reflex method was adopted for evaluation of the anticonvulsant activity of such compounds and screened

by determining their ability to protect the experimental animals against pentylenetetrazole induced convulsion following the method reported by Soaje-Echaque and Lim [24].

Table 5: Anticonvulsant activity of some the newly synthesized compounds

Comp. No.	Dose mg/kg	Protection %	ED ₅₀ mg/kg	ED ₅₀ m mol/kg \pm S.D	Relative Potency M \pm S.D
5 _a	50	33.33	90	0.259 \pm 0.08	0.120 \pm 0.03
	100	50			
	200	100			
5 _b	50	33.3	90	0.249 \pm 0.09	0.124 \pm 0.02
	100	66.6			
	200	100			
5 _c	75	33.3	100	0.266 \pm 0.07	0.116 \pm 0.04
	150	66.6			
	250	100			
6 _a	50	33.3	130	0.312 \pm 0.11	0.099 \pm 0.02
	150	66.6			
	300	100			
6 _b	75	33.3	125	0.290 \pm 0.05	0.106 \pm 0.03
	150	66.6			
	300	100			
Phenobarbitone	75	33.33	100	0.403 \pm 0.12	0.1 \pm 0.021
	125	66.6			
	175	100			

The percent protection, ED₅₀ (mg/kg) and or mmols/kg and the mean of the relative potency \pm S.D. of the test compounds to phenobarbitone as reference drug

Anticonvulsant activity

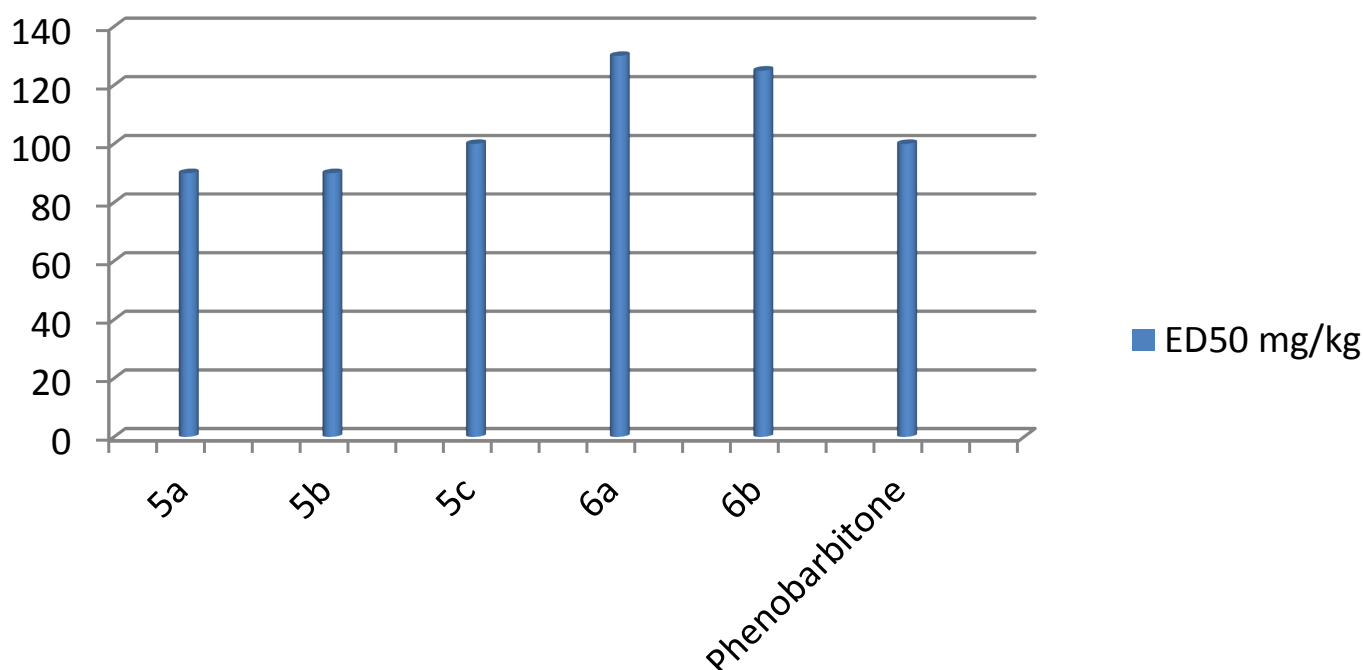


Figure 8. Activity of the new compounds due to its have more than aromatic and heteroatoms which these aromatic ring and heteroatoms.

Preliminary assessment of the anticonvulsant activity

The anticonvulsant activity of some phthalazinediones was assessed in frogs against pentylenetetrazole induced convulsion in comparison to phenobarbitone as a reference drug following the technique of Soaje-Echque and Lim. The test compounds and phenobarbitone were suspended in water with the aid of tween-80 to give suspension with 2% concentration. Pentylenetetrazole was also dissolved in water containing few drops of tween-80 to produce 2% solution. Groups of six Frogs were injected in the dorsal lymph sac with three graded doses of the test compounds or phenobarbitone. After 45 minutes, the animals were injected with the convulsive dose of pentylenetetrazole (320 mg/kg). The animals that showed no tonic convulsion within 60 minutes after pentylenetetrazole administration were considered to be protected. The percent protection, ED-50 (mg/kg) and or mmol/kg and the mean of the relative potency \pm S.D. of the test compounds to phenobarbitone were calculated. The data are presented in Table 5 and Figure 8.

Conclusion

The new compounds which tested as anticonvulsant showed anticonvulsant activity compared with phenobarbitone as reference standard drug. The activity of the new compounds due to its have more than aromatic and heteroatoms which these aromatic ring and heteroatoms have good binding with receptor so showed the anticonvulsant activity.

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