

Medicinal chemistry

Single Polymer-drug Conjugate Carrying Two Drugs for Fixed-dose Codelivery

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Abstract

Fixed-dose combination chemotherapy holds great potential for management of cancer. Thus a drug delivery system which can administer a controlled ratio of several drugs simultaneously and control the drug release at the cancer site is highly desired. In this work, a fixed-dose dual drug loaded polymer micelle is formed by the self-assembly of a single polymer–drug conjugate carrying a combination of drugs. A predetermined ratio of the two drugs can be obtained via a facile and efficient solid-phase synthesis method. MTT assay demonstrated that the polymer micelles are more effective in altering the proliferation rate of MCF-7 tumor cells due to its higher solubility than free drugs. Furthermore, the introduction of the redox-sensitive disulfide linker between the hydrophilic PEG and the hydrophobic drugs facilitates drug release in tumor cellular redox environment and thus enhances the therapeutic effectiveness dramatically.

Keywords: Fixed-dose combination therapy; Polymer micelle; Redox-sensitive; Solid-phase synthesis

Introduction

Due to molecular complexity of many diseases, co-delivery of multiple therapeutic cargos aiming at various targets and displaying different toxicity profiles within the same carrier is becoming increasingly important for enhancing the therapeutic efficacy, overcoming drug resistance, reducing the dose of each agent and reducing side effects [1-4]. A various carriers including liposomes, polymeric micelles, PLGA nanoparticles, dendrimers, mesoporous silica nanoparticles, Janus particles and DNA nanogels have all been adapted for co-delivery [2,4]. Among which, the dual drug loaded polymer micelles usually can be formed via four different ways [5-7]: (1) polymer plus two kinds of free drugs: drugs are loaded during the self-assembly process by including them in the solvent with the polymer. (2) polymer-drug conjugate plus free drug: in this approach a polymer-drug conjugate is formed first, followed by self-assembly and encapsulation of the free drug. (3) Polymer-drug conjugate plus polymer-drug conjugate: two polymer-drug conjugates, each with a single type of drugs, are administered in combination. (4) Single polymer-drug conjugate carrying a combination of drugs: the formation of a chemically mixed micelle containing both types of drugs conjugated to a single polymer. The last two approaches are considered to be promising for fixeddose combination therapy purpose which can direct the system to be synergistic, additive or antagonistic [8,9]. For polymer-drug conjugate plus polymer-drug conjugate, the two populations of conjugates can be mixed in a given ratio to obtain micelles containing a determined ratio of two drugs. While for single polymer-drug conjugate carrying a combination of drugs, more precisely and convenient ratiometric control between the two drugs could be obtained in the initial chemical synthesis process. Also type 4 is the only approach that can guarantee simultaneous delivery of both drugs to the same site of action, and with careful design, can enable synergistic drug effects [10,11]. However, the main obstacle that we would encounter is the difficulty of attaching two different drugs to the same polymer backbone at a controlled fashion.

Here, an unusual and efficient approach was developed to obtain a single polymer–drug conjugate which simultaneously carry two therapeutic agents, acetyl-11-keto- β -boswellic acid (AKBA) and methotrexate (MTX). Importantly, the method is universal to afford a variety of drug ratios but in this study ratio of 1:2 was chosen for simplicity. The rational to choose these two drugs is due to that: AKBA can strongly inhibit tumor angiogenesis and is an anti-inflammatory agent [12]; MTX is a dihydrofolate reductase enzyme inhibitor that is used in the treatment of some types of neoplasias [13]. Functional group is structurally available making another important feature to use them as model drugs. Taking advantage of solid-phase synthesis, AKBA and MTX were first bonded together via the Lysine linkage to yield a dual drug conjugate. The dual drug conjugate is then conjugated to two terminal ends of PEG via EDC chemistry, to obtain polymerdrug conjugate carrying two therapeutic agents. Moreover, for effective therapy, stimuli-responsive disulfide bond was used as a linker between PEG and dual drug conjugate to facilitate tumor relevant glutathione (GSH) triggered release. It is reported that under tumor-relevant reductive conditions, the disulfide-linked PEG could be cleaved and hence the drug release rate could be significantly accelerated from the assembled micelles [14-21].

Experimental Section

Materials and methods

Poly(ethyl glycol) (PEG2000) with carboxyl was purchased from Yare Biotech. Dichloromethane (DCM) and N,N-dimethylformamide (DMF) were dried by refluxing over CaH₂ and distilled prior to use. 9-fluorenylmethoxycarbonyl (Fmoc) and 2-(4,4-dimethyl-2,6-dioxocyclohexylidene)ethyl (Dde) protected amino acids, 2-chlorotrityl chloride (CTC) resin were purchased from GL Biochem (shanghai). O-(benzotriazol-1-yl)-N,N,N',N'- tetramethyluronium hexafluorophosphate (HBTU), thioanisole, trifluoroacetic acid (TFA), N-ethyldiisopropylamine (DIEA) and ninhydrin were purchased from GL Biochem. Dulbecco's modified Eagle's medium (DMEM), fetal bovine serum (FBS), penicillin- streptomycin, trypsin, Dubelcco's phosphate buffered saline (DPBS), 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) were obtained from Gibco

Received July 29, 2014; Accepted September 21, 2014; Published September 24, 2014

Citation: Li Y, Dong H, Li X, Shi D, Li Y (2014) Single Polymer-drug Conjugate Carrying Two Drugs for Fixed-dose Co-delivery. Med chem 4: 672-683. doi:10.4172/2161-0444.1000211

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Invitrogen Corp. 4% paraformaldehyde were purchased from DingGuo Chang Sheng Biotech. Co., Ltd. All other chemicals obtained from Sinopharm Chemical Reagent Company (SCRC) were of analytical grade and were used as received. The dialysis bags were purchased from Spectrum Laboratories Inc. MCF-7 cancer cells were kindly provided by cell center of Tumor Hospital, Fudan University (Shanghai, China).

Synthesis of a AKBA:MTX=1:2 lysine prodrug

Attachment of Fmoc-Lys(Dde)-OH to 2-CTC resin: After swelling in dry DCM (10 ml) for 20 min, the 2-CTC resin (0.3 g, theoretical loading: 1.3 mmol/g, 0.39 mmol) was treated with a solution of Fmoc-Lys(Dde)-OH (0.23 g, 0.5 mmol) in dry DCM (10 ml) and DIEA (340 μ L, 1.95 mmol) at room temperature for 5 h. MeOH (5 ml) was added to cap the free sites, and the reaction mixture was shaken for 15 min. The resin was washed with DMF (5 ml×3), DCM (5 ml×3) and MeOH (5 ml×3), and dried under vacuum for 4 h to obtain Fmoc-Lys(Dde)-OH bound on resin.

Fmoc deprotection: A solution of 20% piperidine in DMF (5 ml×2) was added to the Fmoc-protected and Dde-protected Lysine resin, and the reaction mixture was shaken for 15 min twice. The resin was washed with DMF (10 ml×6). The free amino group was tested with the principle of the ninhydrin reaction to identify the first Lysine being coupled to the resin.

Coupling of the other lysine: Another Fmoc-Lys(Dde)-OH was coupled to the resin. The amino acid (0.26 g, 0.575 mmol) was dissolved in DMF (5 ml), and t hen HBTU (0.4 g, 1.053 mmol) and DIEA (340 μ L, 1.95 mmol) were added. The mixture was added into the resin, and then was shaken at room temperature for 1 h and washed with DMF (5 ml×6). The end of the coupling was controlled by the Kaiser test with the principle of the ninhydrin reaction to detect free amino group.

Fmoc deprotection: A solution of 20% piperidine in DMF (5 ml×2) was added to the Fmoc-protected and Dde-protected Lysine resin, and the reaction mixture was shaken for 15 min twice. The resin was washed with DMF (10 ml×6). The free amino group was tested with the principle of the ninhydrin reaction to identify the first Lysine being coupled to the resin.

Coupling of AKBA-COOH: AKBA-COOH was coupled to the resin. The AKBA-COOH (0.45 g, 0.78 mmol) was dissolved in DMF (10 ml), and then HBTU (0.4 g, 1.053 mmol) and DIEA (340 μ L, 1.95 mmol) were added. The mixture was added into the resin, and then was shaken at room temperature for 1 h and washed with dried DMF (10 ml×6). The end of the coupling was controlled by the Kaiser test with the principle of the ninhtydrin reaction to detect free amino group.

Dde deprotection: A solution of 20% Hydraine in DMF (5 ml×2) was added to the Fmoc-protected and Dde-protected Lysine resin, and the reaction mixture was shaken for 15 min. The above treatment was repeated twice. The resin was washed with DMF (10 ml×6).

Coupling of MTX: MTX was coupled to the resin. The MTX (1.7 g, 3.74 mmol) was dissolved in DMF (10 ml), and then HBTU (0.4 g, 1.053 mmol) and DIEA (340 μ L, 1.95 mmol) were added. The mixture was added into the resin, and then was shaken at room temperature for 1 h and washed with dried DMF (10 ml×6). The end of the coupling was controlled by the Kaiser test with the principle of the ninhydrin reaction to detect free amino group.

Cleavage with TFA: The prodrug-grafted resin was washed with DMF (10 ml×6) and DCM (10 ml×6). Then the prodrug-grafted resin was treated with a solution of TFA and DCM (5:95; 10 ml) for 1 h. The reaction solution was filtered and the filtrates were collected. The

above treatment was repeated twice more. All filtrates were combined and concentrated. Then the filtrates were precipitated with diethyl ether to yield the AKBA: MTX=1:2 lysine prodrug AM_2 . The AM_2 lysine prodrug was washed several times and concentrated by centrifugation to yield the product, which was a light yellow powder.

Synthesis of AKBA-COOH

To a stirred solution of AKBA (512 mg, 1 mmol) in THF (20 ml) maintained at room temperature, thionyl chloride (20 ml) was added dropwise under nitrogen atmosphere for a night. After reaction, the liquid was removed by rotary evaporation. Then glycolic acid (380 mg, 5 mmol) and triethylamine (140 μ L) was dissolved in THF (20 ml), which was then added into the round-bottom flask maintained at 75°C for 8 h. After reaction, the excess THF was removed by rotary evaporation. 10 ml DMF was added into the round-bottom flask, and the mixture was transferred into a dialysis bag (MWCO: 3500 Da) and purified by three-five subsequent dialyzing procedures against deionized water. The precipitated product was lyophilized and stored at -20°C until further use.

Synthesis of H₂N-SS-PEG-SS-NH₂

A stirred solution of HOOC-PEG-COOH (1.0 g, 0.47 mmol) in DCM (25 ml) was combined with 10 N,N'-dicyclohexylcarbodiimide (DCC, 116 mg, 0.564 mmol) and N-hydroxysuccinimide (NHS, 65 mg,0.564 mmol) at RT under nitrogen. After 12 h, the solution was added dropwise into a round-bottom flask containing cystamine (716 mg, 4.7 mmol) dissolved in 5 ml DCM and the reaction was stirred for another 24 h. Following cooling of the mixture to 0°C, precipitated dicyclohexylurea was removed by filtration. The filtrate was evaporated under vacuum and the residue was dissolved in 10 ml of DMSO. The desired product was purified by exhaustive dialysis (MWCO = 1.0 kDa) against deionized water and collected using a membrane filter (450 nm). Following suspension in 10 ml of deionized water, the intermediate H_2N -SS-PEG-SS-NH₂ (0.95 g, yield 81.0%) was lyophilized and stored at -20°C until further use.

Synthesis of polymer-drug conjugate

Polymer–drug conjugate AM_2SS -PEG-SS- AM_2 was prepared following the similar procedure of H_2N -SS-PEG-SS- NH_2 . To a stirred solution of AM_2 (300 mg, 0.70 mmol) in DMF (30 ml) maintained at room temperature and under nitrogen atmosphere, DCC (173 mg, 0.84 mmol) and NHS (96.6 mg, 0.84 mmol) were added. After 12 h a solution of H_2N -SS-PEG-SS- NH_2 (560 mg, 0.23 mmol) and in DMF (10 ml) was added dropwise and the reaction was prolonged for 24 h. The reaction mixture was evaporated under vacuum and the residue dissolved in DMF (10 ml) and diluted with deionized water (10 ml). The resulting mixture was transferred into a dialysis bag (MWCO: 3500 Da) and purified by subsequent dialyzing procedures against deionized water. The precipitate was removed with a 450 nm filter. The purified suspension was freeze-dried giving rise to AM_2 .SS-PEG-SS- AM_2 which was stored at -20°C until further use.

Characterizations

¹H nuclear magnetic resonance (¹H NMR) spectra were recorded with an Avance 500 MHZ spectrometer (Switzerland) using DMSO-d6 as solvent, TMS as standard. UV experiments were conducted on a Cary 50 UV-Vis spectrophotometer (Varian, Ltd., Hong Kong). The molecular weights of AM₂.SS-PEG-SS-AM₂ were performed using the Applied Biosystems 4700 Proteomics (TOF/TOF) Analyzer (Framingham, MA, USA). The UV Nd:YAG laser was operated at a 200 Hz repetition rate wavelength of λ =355 nm. Accelerated voltage was operated at 20 kV under batch mode acquisition control. The solution

was 0.001:1:2 (v/v) trifluoroacetic acid (TFA)-acetonitrile (ACN)-DMF. Mass spectral data were processed using Data Explorer 4.0 (Applied Biosystems). The particle sizes and size distributions of AM2 SS-PEG-SS-AM, micelles were measured by dynamic light scattering (DLS) (DLS, Malvern instruments Ltd., Worcestershire, UK). Morphological observation of AM, SS-PEG-SS-AM, micelles was performed using transmission electron microscopy (TEM) (JSM-200CX, JEOL). A copper grid with a carbon film was used. The copper grid was immersed in a drop of polymer solution for 1 min, and then dried at room temperature. Fluorescence spectra were recorded on a Hitachi F2500 luminescence spectrometer (Hitachi Ltd., Hong Kong). Critical micelle formation concentration (CMC) of AM, SS-PEG-SS-AM, conjugates in aqueous were measured through surface tension technique. A series of aqueous solutions containing different concentrations (from 9.76×10⁻⁴ to 1 mg/ml) of the conjugates was prepared, and the surface tension of each solution was determined individually on an OCA20 contactangle analysis system (Data Physics, Germany) using the pendant drop method. The CMC was evaluated from plots of the static surface tension versus the prodrug concentration. The apparatus was calibrated using the surface tensions of deionized water and pure ethanol.

Micelle formation

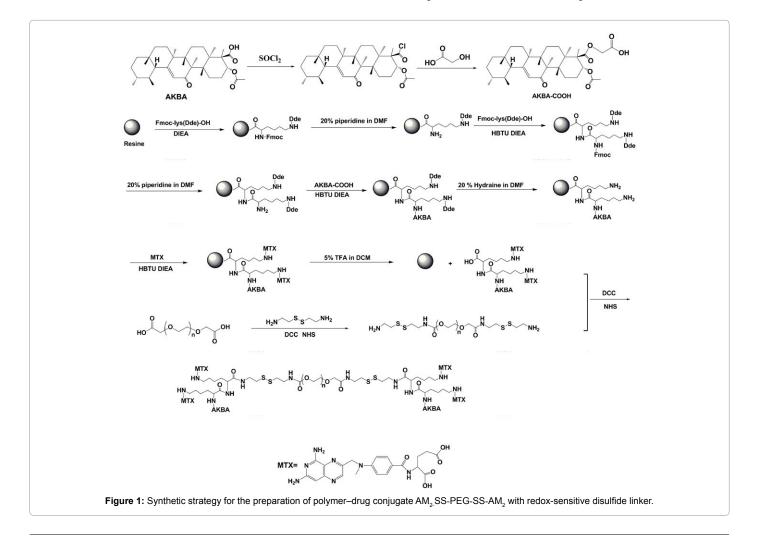
Aqueous suspensions of AM_2 .SS-PEG-SS- AM_2 micelles were prepared by dialysis at RT. Briefly, AM_2 .SS-PEG-SS- AM_2 prodrug (2 mg) was dissolved in DMF at an initial concentration of 0.5 mg/ml and dialyzed for 24 h against 2.0 L of deionized water (MWCO = 3.5 kDa). The water was changed every 6 h.

Degradation of AM, SS-PEG-SS-AM, micelles

 AM_2 .SS-PEG-SS- AM_2 micelle was degraded in the presence of water-soluble reducing agents, glutathione (GSH) in PBS buffer. The concentrations of GSH in mixture solution were set at 10 mM. In a typical procedure, AM_2 .SS-PEG-SS- AM_2 micelle (0.5 mg/ml) was treated with GSH in 250 ml PBS buffer with GSH under at 37°C. The size change of micelles was monitored by DLS measurement at different time intervals.

Release of anticancer agent from micelles and determination of the size change of micelles

The release profiles of AM_2 from AM_2 SS-PEG-SS- AM_2 micelles were studied at 37°C in two different media, i.e. 250 ml PBS buffer (pH 7.4) with 10 mM GSH and neat PBS buffer (pH 7.4) under constant 150 rpm stirring. 10 mg of AM_2 SS-PEG-SS- AM_2 were dissolved in 20 ml neat PBS buffer and formed micelles by ultrasound. Ten milliliter the solution was charged into two dialysis bags (MWCO: 3500 Da) each of which contained five milliliter. Then the dialysis bags were immersed in the dialysis medium. At certain time intervals, 2 ml aliquot of the dialysis medium was withdrawn, and the same volume of fresh media was added, respectively. The sample solution was set at 385 nm for AM_2 . The standard solution of AM_2 lactone was made by dilution



of AM₂ stock solution in PBS buffer. Standard calibration samples were prepared at concentrations ranging from 2.5×10^{-3} to $1.3 \,\mu$ g/ml. The size change of micelles was monitored by DLS measurement before and after release of anticancer agent from polymeric micelles.

Cytotoxicity of AM, SS-PEG-SS-AM, prodrug micelles

MCF-7 cells were maintained in DMEM supplemented with 10% (v/v) FBS and 1% (w/v) of penicillin/streptomycin. For experiments, cells were dissociated from plastic support using trypsin and seeded into 96-well flat-bottomed tissue-culture plates at a density of 5,000 cells/ well. Cells were allowed to attach overnight in a humidified atmosphere of 5% (v/v) CO₂ at 37°C before exposure to 100 µL of single free MTX diluted in culture medium to 1.9-7.5 mg/L, single free AKBA diluted to 1.9-15 mg/L, combination of free AKBA and MTX (ratio:1:2) diluted to 1.9-7.5 mg/L or AM2SS-PEG-SS-AM2 prodrug micelles diluted in culture medium to 1.9-240 mg/L in the presence and absence of 10 mM GSH,. After 24 h incubation, 20 µL of a MTT solution prepared in PBS (5 mg/ml) was added to each well. Subsequently, cells were incubated for additional 4 h at 37°C allowing viable cells to reduce the MTT into purple formazan crystals. 150 µL of DMSO were added to each well to dissolve formazan crystals before absorbance was measured at λ =492 nm using a Multiscan MK3 plate reader (Thermo Fisher Scientific, Waltham, MA, USA). The relative cell viability in % was calculated according to: cell viability = $(OD_{treated}/OD_{control}) \times 100\%$, where OD treated represents the absorbance of treated cells after subtraction of absorbance of control wells containing only cell culture medium.

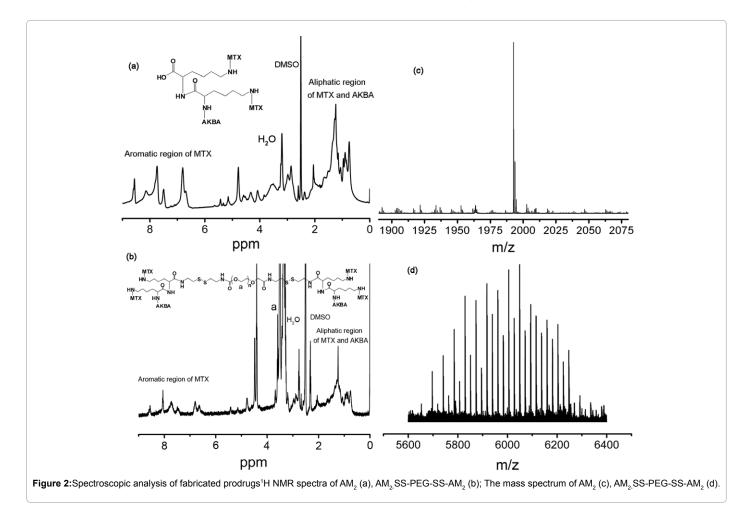
Cellular uptake of AM2 SS-PEG-SS-AM2 prodrug micelles

MCF-7 cells routinely maintained in DMEM supplemented with 10% (v/v) FBS and 1% (w/v) of penicillin/streptomycin was seeded in a 6-well plate at a density of 1×10^5 cells/well. Cells were allowed to attach overnight in a humidified atmosphere of 5% (v/v) CO₂ at 37°C. Before experiments, cells were washed with preheated PBS, pH 7.4 and incubated with 100 µg/ml of AM₂.SS-PEG-SS-AM₂ micelles at 37°C using complete culture medium. After a 24 h incubation period, cells were washed twice with PBS, pH 7.4 and fixed with 4% (w/v) formaldehyde in PBS. The slides were mounted for confocal laser scanning microscopy (Olympus, FV300, IX71, Tokyo, Japan), and intracellular anticancer agent was detected at an excitation wavelength of λ =405 nm.

Results and Discussion

Synthesis of polymer-drug conjugate

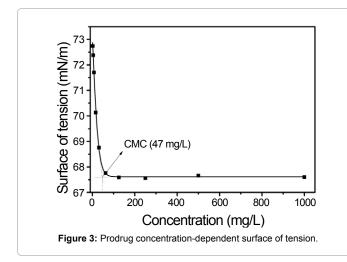
The synthesis process of the polymer–drug conjugate AM₂SS-PEG-SS-AM₂ can be divided into three major parts as illustrated in Figure 1. First, synthesis of Lysine linked AKBA and MTX conjugate in which the molar ratio of AKBA to MTX is 1:2 with a facile solid-phase synthesis method. Solid-phase synthesis is favorable for syntheses of molecules that need to be synthesized in a certain alignment and is a method in which molecules are bound on a bead and synthesized step-by-step in a reactant solution. Compared with normal synthesis in a liquid state, it is easier to remove excess reactant or byproduct from the product. In this method, building blocks 2-(4,4-dimethyl-2,6-dioxocyclohexylidene)



ethyl (Dde) and 9-fluorenylmethoxycarbonyl (Fmoc) protected Lysine Fmoc-lys(Dde)-OH is bound to a solid phase material 2-chlorotrityl chloride (CTC) resin, forming a covalent bond between the carbonyl group and the resin. Then the Fmoc protected amino group of Lysine is deprotected and reacted with the carbonyl group of the next aminoprotected Lysine Fmoc-lys(Dde)-OH. The functional amino groups of Lysine that are able to participate in the desired reaction between building blocks AKBA and MTX can be controlled by the order of deprotection. Fmoc protected amino group is first deprotected and AKBA was added and linked with Lysine. After the conjugation of AKBA, two Dde protected amino group is deprotected and MTX was conjugated. Finally, Lysine linked AKBA and MTX conjugate AM, is cut off from the resin with TFA. It should be noted that although only the ratio 1:2 and model drugs are demonstrated, the strategy can be employed to afford the variable ratio of drugs by controlling numbers of lysines in the conjugate and thus providing the possibility to choose the best drug ratio. For example: one lysine in the conjugate enables the conjugates with 1:1 ratio, while three lysines yield 1:3 or 3:1 ratio.

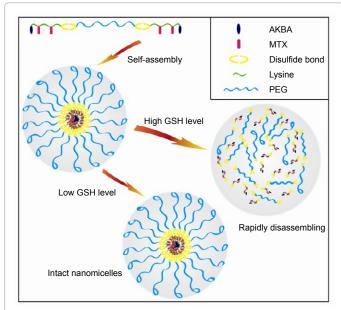
Second, for introducing stimuli-responsive disulfide bond, cystamine was conjugated with HOOC-PEG-COOH to obtain the intermediate H_2N -SS-PEG-SS-NH₂. Third, conjugation of the Lysine linked AKBA and MTX to two terminal ends of H_2N -SS-PEG-SS-NH₂ via EDC chemistry. PEG was chosen to achieve an improved solubility and stability of the drug [22,23]. And moreover, after conjugation with AKBA and MTX, PEG₂₀₀₀ provides a suitable hydrophilic-hydrophobic ratio for the polymer drug conjugate to self-assemble into micelles that enhances therapeutic effectiveness and reduces side effects of the drug payloads by improving their pharmacokinetics.

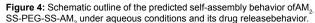
Structural properties of all intermediates and the final product were monitored by ¹H NMR spectroscopy and Proteomics (TOF/TOF) Analyzer. A representative sample of the AM₂ and AM₂SS-PEG-SS-AM₂ is shown in Figure 2a and 2b. Briefly, the peak at δ =3.53 ppm is assigned to the protons of the –CH₂– group in PEG units, the peaks at the aliphatic region from δ =0 ppm to δ =2 ppm, δ =5.12 ppm, δ =5.42 ppm and δ =2.82 ppm are assigned to AKBA, and the peaks at the aromatic region from δ =7 ppm to δ =9 ppm, δ =4.36 ppm, δ =4.79 ppm and δ =3.49 ppm are assigned to MTX. The ratio of AKBA and MTX that is 1:2 is conformed through the calculation results of the aliphatic region of AKBA to the aromatic region of MTX. The successful synthesis of the polymer drug conjugate is confirmed by the aforementioned ¹NMR spectrum. This conclusion is also supported by the mass spectrometry data (Figure 2c and 2d) in line with the anticipation.

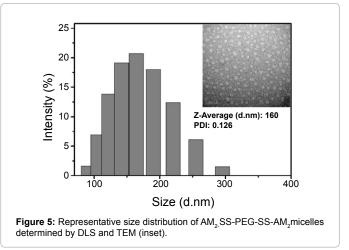


Micellelization behavior of polymer-drug conjugate

The hydrophobic-hydrophilic-hydrophobic block design of AM₂-SS-PEG-SS-AM, implied amphiphilic properties of the synthesized prodrug. Regarding that MTX is fluorescence itself which may influence the measurement of pyrene fluorescence, aggregation behavior of aqueous AM2-SS-PEG-SS-AM2 solutions was measured not by the conventional fluorescent method but through the surface tension method. In surface tension characterization, critical micelle concentration (CMC) was evaluated from plots of the static surface tension versus the prodrug concentration because the surface tension decreases when the amphipathic substance is added [24,25]. Surface tension of the polymer drug conjugate decreases as a function of the polymer concentration, and levels off at around 47 mgL⁻¹ (Figure 3). Consequently, we concluded that micellar assemblies of this novel prodrug were formed at concentrations exceeding CMC. Figure 4 illustrates spontaneous formation of these micellar aggregates. Dynamic laser light scattering (DLS) and transmission electron microscopy (TEM) were used to experimentally assess self-association properties of the synthesized prodrug. DLS clearly revealed the formation of micellar







structures with a mean diameter of 160 nm (Figure 5). TEM confirmed the distinct outline of polymeric aggregates but at a significantly smaller size (50 nm on average, Figure 5). In contrast to DLS determinations, which are performed using aqueous suspension, TEM analysis requires dried samples. The large difference between the experimentally determined average diameter of these micelles by DLS and TEM might be attributed to both the dehydration effect and existing aggregation.

GSH triggering disassembly and in vitro drug release

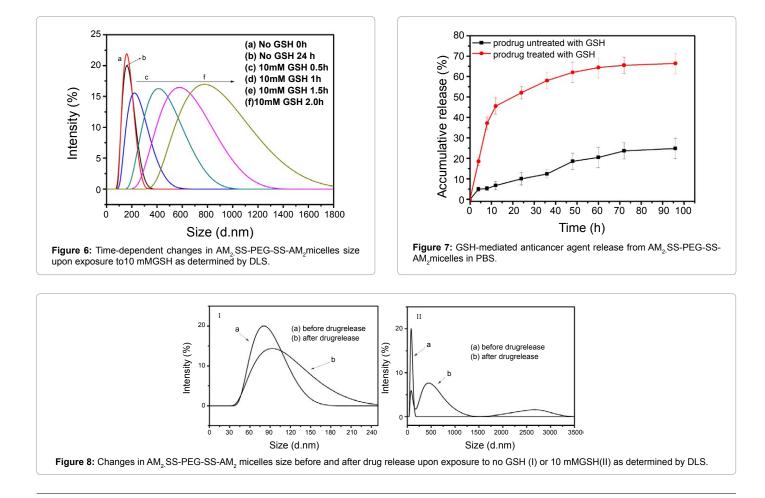
In the design of the polymer-drug conjugate, a disulfide linker is introduced between the hydrophilic PEG and the hydrophobic AM_2 to engineer a redox-sensitive release mechanism for the prodrug. To assess the functionality of this stimulus-induced release mechanism, glutathione (GSH) -induced size changes of the micelles were studied by DLS (Figure 6). There was an insignificant size change over 24 h in the control experiment without GSH. In contrast, following addition of 10 mM GSH, there was a fast alteration in AM_2 -SS-PEG-SS-AM₂ aggregation behavior. Within 2 h, the average micelles size increased from 160 nm to about 1000 nm with concomitant increase of the polydispersity index (PDI) from 0.13 to 0.65. This size increase phenomenon may due to the increasing hydrophobic property of the micelles as a result of PEG shell detachment [26].

Detachment of the PEG diffusion barrier is anticipated to accelerate release of anticancer agent from the prodrug design. Therefore, the in vitro release of the antineoplastic drug was quantified following incubation of AM_2 SS-PEG-SS- AM_2 in the presence and absence of 10 mM GSH (Figure 7). Consistent with results from DLS measurements, only less than 20% of the anticancer agent was observed within 90 h

under non-reducing conditions. In presence of 10 mM GSH, however, the AM₂.SS-PEG-SS-AM₂ micelles exhibited a much faster drug release rate following apparent biphasic kinetics. Within the first 10 h, AM₂ was released about 40%. Subsequent drug release was dramatically reduced with release rate of only 0.3% h⁻¹. This may result from the formation of larger aggregates in the rearrangement process of micelles architecture. Finally the diffusion barrier for reductively cleaved AM₂ is increased and impeded them to reach the bulk solution [21]. Meanwhile, the changes in AM₂.SS-PEG-SS-AM₂ micelles size are consistent with the profile of drug release (Figure 8).

In vitro therapeutical effect

To determine whether AM, released from self-assembled prodrug micelles under tumor-relevant reductive conditions remains pharmacologically active compared to the free agents, cytotoxicity of the micelles was evaluated in the presence and absence of 10 mM GSH using the MCF-7 cell model by the MTT assay. The results summarized in Figure 9 demonstrated that combination of free AKBA and free MTX showed more effective than either single free AKBA or single free MTX in inhibiting the proliferation rate of these tumor cells. The prodrug showed less effective than the free drugs at lower concentration window. This is presumably due to incomplete drug release from prodrug systems. However, at higher concentration window that free drugs can not be soluble, the prodrug system showed effective inhibition on the proliferation rate of these tumor cells. Furthermore, the prodrug was found to be more effective in the presence of tumor-relevant 10 mM GSH concentrations than in the absence of GSH. Meanwhile, inclusion of greater AM, SS-PEG-SS-AM, concentrations in the presence of 10 mM



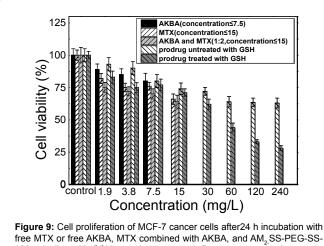


Figure 5: Cell proliferation of MCF-7 cancer cells after24 h included of with free MTX or free AKBA, MTX combined with AKBA, and AM₂,SS-PEG-SS-AM₂ micelles with GSHtreated or untreated. Data are presented as mean \pm SD.

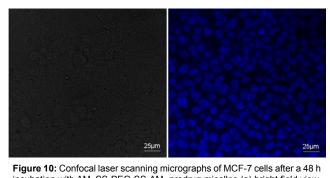


Figure 10: Contocal laser scanning micrographs of MCF-7 cells after a 48 h incubation with AM₂SS-PEG-SS-AM₂ prodrug micelles (a) bright field view, (b) fluorescence view with blue for MTX.

GSH effectively decreased MCF-7 cell viability, suggesting increased intracellular accumulation of pharmacologically active AKBA and MTX. It is conceivable that GSH-stimulated cleavage of the hydrophilic PEG shell altered the release kinetics of AKBA and MTX from micelles and increased the concentration of active anticancer drugs in the cancer cell which finally enhanced the therapeutic effectiveness. This phenomenon has been observed in our previous studies on reduction sensitive drug delivery systems [19,21,26]. Confocal laser scanning microscopy results of MCF-7 cells treated with micelle prodrugs clearly demonstrated the effective intracellular presence of AKBA and MTX in MCF-7 cells after a 48 h incubation period (Figure 10).

Conclusion

A facile and efficient solid-phase synthesis method was used to synthesize a single polymer–drug conjugate which simultaneously carry two therapeutic agents via a biodegradable linker. The synthesized amphiphilic polymer–drug conjugate can self-assemble into micelles with sizes around 100 nm. The micelle prodrug showed more effective in altering the proliferation rate of MCF-7 tumor cells at high concentration due to its higher solubility than free drugs. Under tumorrelevant reductive conditions, the micelles exhibited a GSH triggering disassembly behavior and the release rate, the therapeutic effectiveness of the antitumor agent was dramatically enhanced.

Acknowledgements

This work was financially supported by 973 program (2013CB967500) and

National Natural Science Foundation of China (51173136, 51473124 and 21104059), the Fundamental Research Funds for the Central Universities" (2013KJ038), and "Chen Guang" project founded by Shanghai Municipal Education Commission and Shanghai Education Development Foundation.

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