

Novel Microstructure Mechanical Activated Nano Composites for Tissue Engineering Applications

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Abstract

Nowadays, tricalcium phosphate TCP ($\text{Ca}_3(\text{PO}_4)_2$) that belongs to the group of calcium phosphate (CaPs) biomaterials as a hot topic of research for bone tissue repair applications. In this investigation, TCP powder was synthesized by mechanical activation (MA) method as a solid state process. The mixture was composed of a blend of pure calcite (CaCO_3) and silica amorphous (SiO_2) powder with 57% wt. and 43% wt., respectively. Then, the powder mixture milled by high energy ball mill with ball to powder ratio (BPR) 15:1 and rotation speed 600 rpm for 10 h. After then, the materials milled heated at three temperatures 900°C, 1000°C and 1100°C for 2 h in muffle furnace at the air atmosphere. X-ray diffraction (XRD), scanning electron microscopy (SEM) and BET technique performed on heated powders to characterize. According to XRD results, the patterns show that the phase TCP was just appeared in the mixture milled for 10 h. In addition, based on modified Scherrer equation the TCP crystalline size was determined 40 nm. In fact, the present investigation indicated that TCP powder was composed of Nano-crystallite structure, 30-40 nm, can be prepared by MA method at 900°C or 1000°C to use as a new biomaterials for tissue engineering application.

Keywords: Synthesis; TCP; Ball Milling; Mechanical Activation; Nanostructure

Introduction

Calcium phosphates (CaPs) biomaterials are widely used as substitutes for auto genus bone grafts when bone reconstruction is considered [1-2]. CaPs such as hydroxyapatite [HA ; $\text{Ca}_{10}(\text{PO}_4)_6(\text{OH})_2$] and Tri-calcium phosphates [TCP; $\text{Ca}_3(\text{PO}_4)_2$] are remarkably biocompatible, provoke little if any inflammatory response, and generally are accepted to be bioactive and Osseo conductive when implanted into bone defects [3]. In orthopaedics a variety of biomaterials has been investigated and widely applied in bone-defect management, and skeletal tissue engineering, with many satisfactory clinical outcomes [4]. HA and TCP are bioactive, these biomaterials that have similar composition, structure close to mineral bone of human have good bioactivity [5,6]. HA-TCP based biomaterials can guide bone tissue growth, facilitate bone formation have been used clinically as bone substitutes. The bio activate material serves as a scaffold for new bone formation and increase the in growth of osteoprogenitor cells, capillaries, and per vascular tissue from the recipient bed. Biomaterials should participate in the continuous process of bone remodeling, and the implanted biomaterial should be gradually replaced at a rate comparable to that of in growth of newly formed bone [7]. Although, excellent bioactive ceramics such as hydroxyapatite ($\text{Ca}_{10}(\text{PO}_4)_6(\text{OH})_2$) [8-10], diopside (CaMgSiO_6) [11], akermanite ($\text{Ca}_2\text{MgSi}_2\text{O}_7$), hardystonite ($\text{Ca}_2\text{ZnSi}_2\text{O}_7$) [12], and forsterite (Mg_2SiO_4) [13] bonds with hard tissues as well. Among these bioceramics, TCP is one typical bioactive ceramic which has shown great potential for bone tissue application. In the present investigation, the TCP powder was composed of Nano-crystallite structure can be prepared by mechanical activation (MA), solid state process, to use as a new biomaterials for medical and dental purposes [6-10].

Materials and Methods

Amount of raw materials required for synthesis of 10 g TCP sample was 5.7 g magnesia (MgO) and 4.3 g silica. The raw materials milled by high energy milling with ball-to powder ratio 10:1 and rotation speed 600 rpm. The mixture milled was heated for 3 hours at three

temperatures (900°C, 1000°C and 1100°C) in the muffle furnace at the air atmosphere. Phase structure analysis was carried out by X-ray diffraction (XRD) (Philips X'Pert-MPD diffract meter with Cu K_α radiation ($\lambda=0.15418$ nm) over the 2θ range of 10–90). The obtained experimental patterns were compared to the standards compiled by the Joint Committee on Powder Diffraction and Standards (JCDPS) which involved card (00-090169) for TCP. The crystallite size of prepared powders was determined using XRD patterns and modified Scherrer equation [14]. Scanning electron microscopy (SEM) analyses evaluations were performed using a LEO 435 VP to investigate the morphology. Samples coated with Au by sputter spraying, low vacuum and 100-120 V accelerating voltage. The specific BET surface area of TCP powder has been done by Kelvin B100.

Modified Scherrer equations

The modified Scherrer equation can provide the advantage of decreasing the sum of absolute values of errors, $\sum(\pm\Delta \ln\beta)^2$, and producing a single line through the points to give a single value of intercept $\ln K\lambda/L$. At this sample, Figure 1, the linear regression plot is obtained as $y = 3.355X - 5.540$. This is equivalent to $\ln \beta = \ln (1/\cos\theta) + \ln (k\lambda/L)$ (Eq. 1). From this line, the intercept is -5.540 and $e^{-5.540} = K\lambda/L$ and $L=35$ nm. So, TCP crystallite size average is 40 nm [5] (Figure 1).

According to the same way, the TCP phase crystallite size for the samples heated at 1000°C and 1100°C are 40 and 50 nm, respectively.

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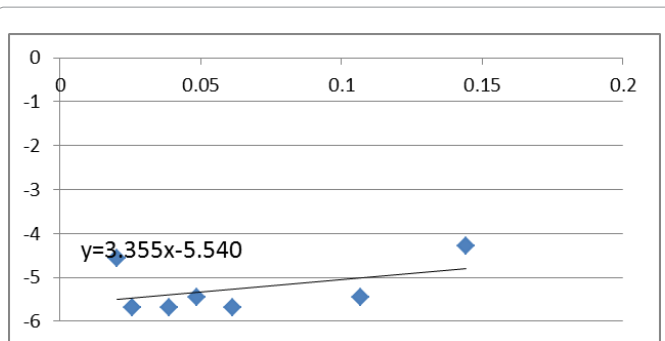


Figure 1: Plot of $\ln\beta$ vs. $\ln(1/\cos\theta)$ of TCP sample heated at 900°C for 2 h.

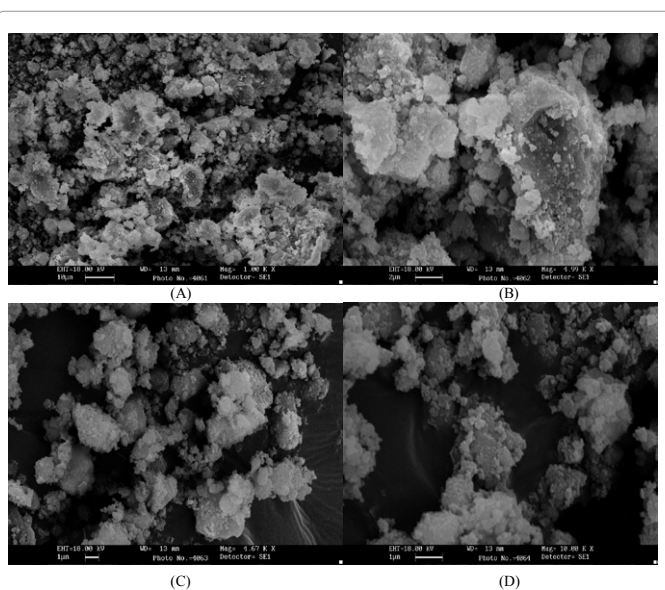


Figure 2: SEM micrograph of the materials milled and then heated at 900°C 1000C, 1100°C for 2 hours.

So, the average crystallite size of TCP phase obtained is approximately 30-50 nm.

Results

SEM micrographs

Figure 2 show the SEM micrograph the materials milled before and after heating. Figure 2 shows the SEM micrographs of the 2 h ball-milled powder after annealing at 1000°C for 1 h. SEM observations showed that TCP has non-spheroid crystals; however, the crystal sizes of TCP is in the order of microns, and some particles are spherical whereas others have sharp shapes (Figure 2a-2c). The agglomerated particles are composed of very fine particles (Figure 2).

XRD Results

Figures 3 and 4 show the XRD patterns of the materials mixture milled heated at three temperatures.

BET results

The specific surface area of the prepared powder was calculated from the N_2 gas adsorption isotherms using the multipoint BET

technique The average particle size of the prepared powder, assuming that the particles synthesized were spheroid, was calculated as shown in Equation 2.

$$D = 6000 / (S_{BET} * d) \quad (2)$$

Where d and D are true density (g/cm^3) and the average particle size (micron) of materials mixture milled, respectively. The specific surface area determined by BET (Figure 5b) was $4.5 m^2/g$. Therefore, the particles size was estimated about 530 nm. In fact, the powder mixture is micron size (Figure 5).

Time and energy transfer

The minimum synthesis ball milling time or best operational performance of milling process is given by equation 3:

$$t = B * m_{ch} * f \quad (3)$$

Which B is constant depending on density of balls, elastic and inelastic collisions as well as the heat released by the reactions between the raw materials; m_{ch} is the mass of the powder charge which is determined by ball to powder weight ratio (BPR); and f is the cost function of the problem so that by minimizing the f , the synthesis time is also minimized. Therefore, in such a scenario, all parameters including B and f play a fundamental role and they have to be optimized

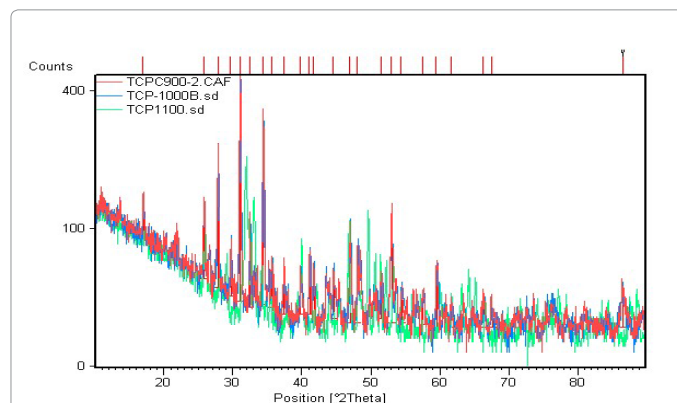


Figure 3: Comparison of XRD pattern of the materials mixture milled and then heated at three temperatures for 2 h.

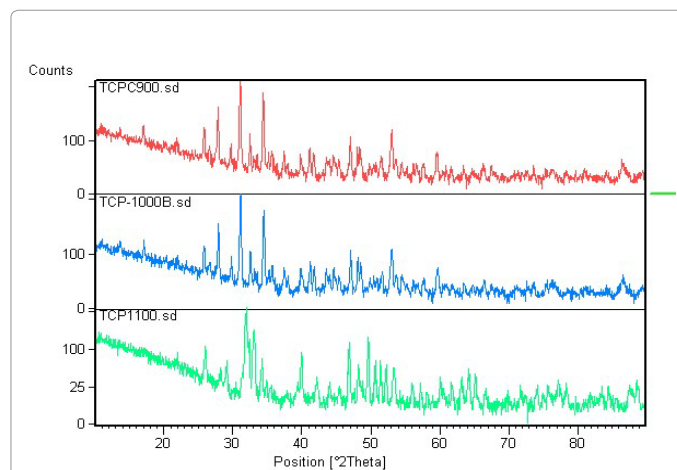


Figure 4: Intensity of sharp peaks of synthesis powder is between 25- 55°.



Figure 5: Equipment used in fabrication of TCP (a) XRD, (b), BET and (c) SEM equipment's in advanced materials research Centre of IAUN University.

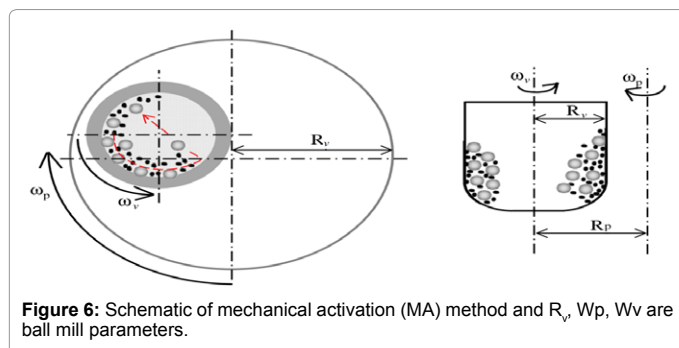


Figure 6: Schematic of mechanical activation (MA) method and R_v , ω_p , ω_v are ball mill parameters.

in order to achieve the minimum synthesis time or best operational performance of ball milling process. This calculation also proved that the optimum time to produce TCP by MA method (Figure 6) is 10 h ball milling.

Discussion

According to the XRD patterns, TCP phase was synthesized in all the samples heated at three temperatures. Based on the previous studies, TCP phase was synthesized at above 1000°C. But, in this method TCP phase have been achieved at 900°C by using mechanical activation. The samples heated at 900°C and 1000°C have same patterns. Whereas, the pattern belong to the sample heated at 1100°C has somewhat right shift in comparison to another patterns. Therefore, the sample heated at 900°C is the best sample containing TCP phase. Bio-ceramics and especially TCP plays a key role as bone tissue engineering and biomaterials in clinical fields. However, the use of conventional materials like HA which is economical cost is limited because of their essential properties. Bovine bone mimicked with TCP structure and produced a proper bioactive coating. In spite of the fact that in other

studies, researches working on different types of HA derive from seashells (sHA) and eggshells (eHA), to compare the bone regeneration ability. Some other types of HA is that, biphasic calcium phosphate (BCP) consisting of HA having excellent bioactivity and BCP having high biodegradability [15-17].

Conclusions

In conclusion, TCP phase have been synthesized at 900°C by materials mixture MA method. Whereas, based on the previous studies, TCP phase was synthesized at above 1000°C. However, the use of conventional materials like HA and TCP which is economical cost is limited because of their essential properties.

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