Nitric Oxide: Friend or Foe in Cancer Chemotherapy and Drug Resistance: A Perspective

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Abstract
A successful treatment of cancers in the clinic has been difficult to achieve because of the emergence of drug resistant tumor cells. While various approaches have been tried to overcome multi-drug resistance, it has remained a major road block in achieving complete success in the clinic. Extensive research has identified various mechanisms, including overexpression of P-glycoprotein 170, modifications in activating or detoxification enzymes (phase I and II enzymes), and mutation and/or decreases in target enzymes in cancer cells. However, nitric oxide and/or nitric oxide-related species have not been considered an important player in cancer treatment and or drug resistance. Here, we examine the significance of nitric oxide in the treatment and resistance mechanisms of various anticancer drugs. Furthermore, we describe the significance of recently reported effects of nitric oxide on topoisomerase and the development of resistance to topoisomerase-poisons in tumor cells.

Keywords: Nitric oxide; Chemotherapy; Anticancer drugs; Topoisomerases; Drug resistance

Introduction
The emergence of cancer cells resistant to chemotherapy drugs has been a major problem in successful treatment of human tumors in the clinic [1-4]. A flurry of research both in the laboratory and in the clinic have identified several mechanisms to explain this resistance [1-4]. One of the most important observations has been the identification of p-170 glycoprotein (pgr), which was overexpressed in resistant tumor cells following drug treatment. Subsequently, it was shown that p-170 glycoprotein was involved in removing drugs from the cancer cells in an energy-dependent manner. Furthermore, it was soon discovered that this protein was also involved in the removal of many drugs that were structurally very different from one another. Rapid progress has identified many other mechanisms of cancer drug resistance, including mutation of certain genes (e.g., p53, topoisomerases), decrease in activity of proteins (e.g., topoisomerases) and modification of drug metabolism and detoxification (phase I and II) enzymes [5-7]. While significant progress has been made towards understanding the mechanisms of cancer drug resistance and we can now adequately explain mechanisms of resistance to ionizing radiation and many commonly utilized chemotherapeutic drugs in the laboratory, significant problems remain in the clinic, as modulation of these pathways and enzymes has not been very successful in curing cancers or alleviating human suffering.

Cancers are extremely complicated and involve many interrelated pathways; no single mechanism of resistance is expected to explain all of them. Recent studies suggest that nitric oxide (NO) may play an important role in the treatment of human cancers. NO is a short-lived free radical molecule that easily diffuses in tissues. It is an important cellular messenger and plays a role in many important physiological functions, including vasodilatation, apoptosis, and the innate immune response. It is believed that many of the actions of NO as a signalling molecule results from its interactions with the heme moiety of the soluble guanylate cyclase, causing the activation and production of second messenger cyclic GMP [8,9]. Recent studies further suggest that many of the actions of nitric oxide results from formation of its reactive metabolites (NO+, N2O3, -OONO) and modification of drug metabolism and detoxification (phase I and II) enzymes [5-7]. While significant progress has been made towards understanding the mechanisms of cancer drug resistance and we can now adequately explain mechanisms of resistance to ionizing radiation and many commonly utilized chemotherapeutic drugs in the laboratory, significant problems remain in the clinic, as modulation of these pathways and enzymes has not been very successful in curing cancers or alleviating human suffering.

Nitric oxide is also implicated in cancer cell killing, cancer progression and metastasis and poor survival [13-17]. A poor clinical outcome is believed to result from NO-induced rapid tumor growth, resulting in hypoxia, nutrient deprivation, poor drug delivery, and selection for drug-resistant tumor cells. In vivo, NO is formed from L-arginine by nitric oxide synthase (NOS). Three forms of NOS have been identified, including neuronal (nNOS), endothelial (eNOS), and a Ca2+-independend inducible isofrom (iNOS). High expression of iNOS and increased production of NO have been described in many human tumors, including breast, prostate and colorectal cancers [17-20]. Increased and continuous generation of NO plays a significant role in the regulation of cancer cell progression and carcinogenesis. Hibbs et al. [21,22] and Stuehr and Nathan [23] have shown that co-culturing of leukemia L1210 cells with activated peritoneal macrophages results in the inhibition of L1210 cell proliferation which is correlated with nitrite formation. Inhibitors of NOS (L-NAME) and myoglobin, a scavenger of NO, were effective in inhibiting cell killing actions of the activated macrophages, suggesting that NO is responsible for the killing actions of activated macrophages towards cancer cells.

It is believed that DNA damage and apoptosis induced by NO in tumor cells cause tumor cell death; studies are needed to explore processes to deliver the highest concentrations of NO in tumor cells using NO-donors as single chemotherapeutic agents or in combination with other chemotherapeutic agents. While significant progress has been made in the last decade, the role of NO/NO-related species remains poorly understood in chemotherapy and drug resistance. Here, we review the role of NO/NO-related species in cancer chemotherapy.

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Received September 29, 2016; Accepted October 26, 2016; Published October 28, 2016


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Nitric Oxide: A Friend in Chemotherapy

Nitric oxide as a single agent

Because 'NO is a gas and forms many reactive metabolites in the presence of molecular oxygen, it is difficult to deliver and study the biological effects of 'NO in tumor cells in vivo. To overcome these problems and to deliver 'NO in tumor cells in vivo in a consistent manner, many chemicals have been synthesized that release 'NO at physiological pH (at various half-lives from minutes to days) or undergo bio-activation in vivo to release nitric oxide. Maragos et al. [24] were the first to report that 'NO released from nitric oxide-donors was inhibitory to A375 human melanoma tumor cells in vitro. A structure activity relationship study indicated that compounds that released 'NO slowly were more toxic to cells than those compounds that released 'NO rapidly, suggesting that sustained exposure to 'NO for a longer time is more damaging to tumor cells. Thus, both the amount and exposure time were critical for cell cytotoxicity. Several NO-donors have now been evaluated as anticancer agents in various tumor cell lines, including organic nitrite, glyceryl nitrite, nitrosothiols and dazieniumdiolates. Several excellent reviews are already available highlighting the importance of NO-donors as anticancer agents [15,25-29]. Therefore, these are not focused on in detail in this review.

Wink et al. [30] have shown that certain NO-donors sensitized cisplatin cytotoxicity in Chinese hamster ovary cells. High concentrations of 'NO released from these NO-donors have been found to induce apoptosis and inhibition of cell growth. These NO-donors were found to be effective in inducing growth inhibition in both wild-type p53-containing and p53-deficient cell types. However, it is interesting to note that some differences in toxicity were observed, i.e., p53-deficient SW620 cells were more sensitive than HCT116, wild-type p53-containing cells. In a series of cell lines derived from colon cancer, the status of cyclooxygenase and/or activity was not important for 'NO-induced cell death [31].

Because 'NO and/or reactive species derived from it react with cellular components and cause damage to both normal cells and tumor cells, it is necessary to deliver NO-releasing toxic compounds directly to tumor cells in vivo (target specificity). Various prodruge-containing 'NO moieties have therefore, been synthesized. One such approach utilized synthesis of esterase-sensitive nitric oxide donors of the dazieniumdiolates [32]. These compounds were found to be active against leukemia cells, but only after releasing 'NO via activation by certain tumor cell esterases.

In keeping with the idea of selectively destroying tumor cells by 'NO without affecting normal cells, Shami et al. [33] designed and synthesized various produgs for NO-donors containing O'-aryl dazieniumdiolates. One such compound JS-K (O'-(2,4-dinitrophenyl)-1-[(4-ethoxycarbonyl)piperazin-1-yl]dazien-1-ium-1,2-diolate) was found to be an extremely active anticancer agent against human HL60 leukemia cells in vitro and reduced the growth of subcutaneously implanted HL60 cells in nude mice [33]. JS-K is active against human non-small cell lung cancer cells [34] and induces apoptosis in human multiple myeloma cells both in vitro and in vivo [35]. It was found that JS-K was activated following reactions with glutathione to produce 'NO in vivo catalyzed by glutathione S-transferase [33]. Shami et al. [33,36-38] have extensively characterized interactions of JS-K with glutathione/glutathione S-transferase and have found that the GST-µ form is the major enzyme responsible for releasing 'NO in vivo. Targeting of tumors by 'NO was also achieved with glycol-S-nitrosothiols (sugar-SNAP) that resulted from enhanced uptake of saccharides in tumor cells and delivering 'NO directly to tumor cells [34]. Reynolds et al. [29] have reported synthesis of polysaccharide-based dextran thiomers which contained covalently attached NO-donors. They found that these NO-donor compounds were stable and quantitatively released 'NO under normal physiological conditions. While these compounds were found to be extremely effective in killing bacteria, no data is currently available about their cytotoxic profiles in human tumor cells either in vitro or in vivo.

Nitric oxide in combination with antitumor agents

Since the early observations of Wink et al. [30] showing that certain NO-donors enhanced the cytotoxicity of cis-platin in Chinese Hamster V79 lung fibroblast cells, extensive research has suggested that the delivery of 'NO via NO-donors has been beneficial and/or synergistic with various anticancer drugs in killing tumor cells in vitro and in animal models [15,25-29]. Interestingly, not all nitric oxide donors are effective in this sensitization of cytotoxicity as it was found to depend upon both the flux of 'NO and the reactive species formed. Furthermore, the sensitization to cis-platin cytotoxicity was short-lived [30]. Since these important observations, a significant amount of research has been carried out showing sensitization to cis-platin and other Pt-based drugs by NO-donors in tumor cells [28,39-41] and in vivo in nude mice. An NO-donor derived from aspirin (NCX-4016) was also effective in sensitizing cisplatin cytotoxicity in both sensitive and cisplatin-resistant cell lines [40]. It is believed that decreases in cellular glutathione played a significant role in the mechanisms of this sensitization [40]. Combinations of NCS-4040, an NO/NSAI/1D methylglycine-DNA methyltransferase activities in the tumor cells [44].

It is important for 'NO delivery in vivo to be selective; one such approach involves gene therapy with iNOS in tumor cells. A successful therapy has been reported using gene transfer with iNOS and cisplatin against prostate, colon and sarcoma tumors in vivo [43]. Targeted 'NO has also been shown to significantly enhance the sensitivity of carmustin (BCNU) and temozolomide against glioma cells, and which was suggested to result from altered expression of O6-methylguanine-DNA methyltransferase in tumor cells [44].

A number of studies have shown that certain NO-donors when combined with doxorubicin are also effective against tumors [45,46]. Furthermore, certain NO-donors e.g. DETA-NO, and nitroglycerine (GTN), were effective in enhancing doxorubicin cytotoxicity and reversing hypoxia-induced resistance to doxorubicin [47]. 'NO was shown to enhance doxorubicin cytotoxicity in a mouse model against prostate tumors as well as against breast MDA-MB-231 cells in a spheroid model [46]. This is in contrast to our studies with PPNO, an NO-donor, which had no effect on doxorubicin cytotoxicity against human breast MCF-7 and colon HT-29 cell lines in vitro [48]. We also found that 'NO did not modulate the toxicity of doxorubicin against human melanoma cells [49]. This suggests that PPNO, which has a short half-life, is not an effective modulator of doxorubicin cytotoxicity.

Because 'NO modulates the activities of various anticancer drugs under hypoxic conditions, the application or use of NO-donors in enhancing chemotherapy in solid tumors is becoming increasingly important. Solid tumors in general are hypoxic due to poor intratumor blood flow. Under hypoxic conditions, tumors have been shown to accumulate hypoxia-inducible factors (HIF) due to lack of prolyl hydroxylase-dependent ubiquitination and degradation of HIF1-a [50,51].
While studies show that 'NO delivered via NO-donors sensitizes/enhances cis-platin activity both in vitro and in vivo, several observations indicate opposite effects of 'NO in the development of resistance to cis-platin in several human tumor cells, including lung, colon and melanoma [52,53]. It has been shown that 'NO nitrosylates bcl2 protein, which inhibits ubiquitination and proteosomal-dependent degradation of bcl2 protein, resulting in its stabilization [52]. This stabilization then inhibits apoptosis and induces cis-platin resistance. Furthermore, it has been shown that both intracellularly and exogenously generated 'NO (delivered via an NO-donor) in human melanoma tumor cells induces significant nitrosylation of various cellular proteins, including caspase-3 and prolyl-hydroxylase-2 [53]. Prolyl-hydroxylase-2 is responsible for the hydroxylation of hypoxia inducible factor-1α (HIF-1α), leading to the degradation of HIF-1α via a proteosomal pathway. Alternation/inhibition in the activities of these enzymes is believed to cause induction of cis-platin resistance in melanoma tumor cells.

These opposite observations should be properly addressed. Are there large differences in the activity of 'NO due to concentration differences of NO-donors used in these experiments? Does the selection of NO-donors or use of certain tumor cell lines influence the results? The observations that endogenously formed 'NO in human lung and melanoma cells causes resistance to cis-platin would suggest that low levels of 'NO (as tumor cells in general do not generate large amounts of 'NO as formed in macrophages from the induction of iNOS) is sufficient to induce resistance to cis-platin. However, 'NO is continuously generated in these tumor cells, which may act differently than a bolus injection of 'NO when delivered via an NO-donor only for a short duration. Whatever the reasons they must be resolved with further studies.

A part of the conflicting effects of 'NO on sensitization or emergence of drug resistant cells may simply be due to the known dose-dependent dual effects of 'NO in cancer cell biology [54-56]. Nitric oxide can either stabilize or induce degradation of HIF-1α [57,58]. It has been observed that 'NO can nitrosylate HIF-1α directly, resulting in its stabilization, and that stabilization is independent of the prolyl hydroxylyase-dependent mechanism. It has been shown that the effects of 'NO on HIF-1α are biphasic: At high 'NO concentrations (>1.0 µM) HIF-1α is stabilized, and at lower concentrations (<400 nM) HIF-1α is rapidlydestroyed [57]. It would appear then that depending upon the concentrations of 'NO (thus, the concentrations of NO-donors) used to evaluate the cytotoxicity of cis-platin or doxorubicin in tumor cells, one can get either sensitization or resistance to the drug.

Nitric Oxide: A Foe in Cancer Chemotherapy

While the role of 'NO and its related species in cancer drug resistance is not known, there now exists some evidence that 'NO and its related species plays a significant role in the development of drug resistance and in the selection of resistant tumor cells in vivo. Topoisomerase-based resistance induced by nitric oxide

A number of clinically active drugs interact with topoisomerases (I and II), which are nuclear enzymes responsible for maintaining the topology and functions of DNA [59-63]. It has been shown that interference with the functions of these enzymes results in inhibition of cell synthesis and ultimately cellular death. Resistance to these topoisomers results from mutation of the topo gene, and/or decreases in activity of the proteins [59-63]. We have found that 'NO nitrosylates both topoisomerase I and II in human breast MCF-7 tumor and colon tumor HT-29 cells when treated with PPNO, an NO-donor [48,64]. We have recently reported that nitrosylation of topoisomerase I leads to proteosomal degradation with time and causes significant resistance to camptothecin, a topo I-specific drug [64]. Interestingly, this nitrosylation of the protein had no significant effect upon the activity of the protein nor did it decrease topo I-dependent DNA damage in tumor cells. Resistance to camptothecin was found to result from 'NO -induced wtp53-dependent upregulation/stabilization of bcl2 protein in MCF-7 cells, as no resistance was observed in HT-29 tumor cells (mutant 53-containing cells) [64].

We have also found that 'NO related species react directly with etoposide (a topoisomerase II-poison) or etoposide-derived radical. The products of this reaction were found to be significantly less toxic to tumor cells than the parent drug [65]. Furthermore, we have recently shown that 'NO also causes nitrosylation of topoisomerase II-SH functional groups in cells [48]. Unlike, topoisomerase I, nitrosylation of topoisomerase II leads to a decrease in its catalytic and relaxation activities both in vitro and in tumor cells [48]. An 'NO-induced decrease in topoisomerase II activity leads to a decrease in DNA damage and induction of significant resistance to etoposide, m-AMSA and XK469 in MCF-7 breast tumor cells. It is interesting to note that while 'NO (or its related species) was able to nitrosylate active -SH functions of both topoisomerase I and II similarly, only the activity of topoisomerase II was significantly compromised, resulting in less DNA damage. While the exact reason (s) for this is not known, we have recently found that 'NO significantly inhibits ATPas activity in general (manuscript in submission; there is no ATP binding site in topoisomerase I and it does not require ATP for its activity). Thus, this inhibition of ATPase activity results in a decrease in topoisomerase II activity and decrease DNA damage as observed in our studies.

Our observations that 'NO inhibits decatenation reactions of topoisomerase II without inducing DNA cleavage [48] present another possible mechanism for 'NO-induced tumor cell killing. Such compounds are known as catalytic inhibitors and including ICRF-193 and related bisdioxopiperazines. These drugs inhibit double-strand break formation, and activate the decatenation checkpoint [66,67]. Impaired decatenation checkpoints have been reported in various bladder and lung cancer cell lines. Nakagawa et al. [68] have shown that certain lung tumor cell line with deficient decatenation checkpoints were highly sensitive to ICRF-193, suggesting that it is possible to design a catalytic inhibitor as an anticancer agent to target specifically altered tumors. Thus, based upon our observations that 'NO is a topo II catalytic inhibitor, an appropriate nitric oxide donor can be designed as an anticancer drug for appropriate tumors in the clinic.

Miscellaneous mechanisms of resistance induced by nitric oxide

It has been reported that mammalian cells incubated with excess 'NO accumulate p53 protein, but concomitantly this p53 loses its capacity for binding to its DNA consensus sequence. Chaizote-Aubert et al. [69,70] have shown that treating wild-type p53-containing MCF-7 cells with S-nitrosoglutathione resulted in the nitration of p53 protein. Furthermore, treatment of MCF-7 tumor cells with S-nitroso-N-acetyl-DL-penicillamine (2 mM to 5 mM) or S-nitroso-glutathione (1 mM to 2 mM) resulted in a rapid accumulation of p53 protein in the nucleus. This p53 protein, however, possessed a significantly decreased activity of specific DNA binding, suggesting modification/alteration of the protein. In contrast, lower concentrations of NO-donors (0.25 mM to 0.5 mM) stimulated p53 accumulation as well as its DNA binding activity. These results suggest that excess 'NO produced in inflamed tissues or in tumor cells could lead to impaired p53 functions that may
play a significant role in selection of resistant tumor cells in vivo or affect DNA repair in tumor cells.

Studies have shown that NO modulates the activity of various DNA repair proteins [71-73], most notably DNA-PKcs, an enzyme responsible for repairing both single and double-strand breaks [73]. This NO-dependent upregulation of DNA-PKcs activity afforded significant protection from several DNA-damaging agents, including X-ray radiation, cis-platin, doxorubicin and bleomycin [73]. Xu et al. [73] have shown that both endogenously and exogenously generated NO induced upregulation of DNA-PKcs in several human tumor cells, including lung A549 and glioma A172 tumor cells, and ultimately caused resistance to ionizing radiation and cis-platin. Induction of DNA-PKcs by NO was clearly shown to cause increased repair of DNA-double strand breaks in these tumor cells. Liu and Xu [60] have further shown that endogenously generated NO in MCF-7 cells also upregulates other DNA repair proteins, e.g., topoisomerase II and Rad2, in addition to DNA-PKcs and HIF-1α.

Studies from various laboratories indicate that NO nitrosylates certain drug target proteins, altering their activity and/or inducing their degradation [11,52,74,75]. These events ultimately can induce resistance to drugs in vitro and in the clinic. NO has been shown to induce cis-platin resistance in human melanoma tumor cells by nitrosating bcl2 protein, causing its stabilization, and inhibiting cis-platin-induced apoptosis in tumor cells [52]. In this regard, Wongvaranon et al. [76] have shown that long term exposure (7-14 days) to NO (via an NO-donor) resulted in resistance to cis-platin, doxorubicin and etoposide in lung NCI-H292 tumor cells. It was shown that the induction of resistance to these drugs was due to an increase in cellular carnitine-1 and bcl2 proteins and up-regulation of activated protein kinase B (AKT). NO treatment did not result in a stable drug-resistant phenotype and resistance was reversible when cells were cultured in the absence of NO [76].

Glynn et al. [17] have shown that increased NOS2 (iNOS) in women with ER negative breast tumors may be linked to poor survival in the clinic. Heinecke et al. [77] have further shown that stimulation of NOS2, such as that found in tumors contaminated by circulating macrophages or during hypoxia, induces the upregulation of NO-targeted biomarkers, e.g., IL-6, IL-8, calcium binding protein S-100A8, and tissue inhibitor matrix metalloproteinase-1. These bio-markers have been suggested to promote cell survival and cancer progression. Using MDA-231 breast cancer cells as a model in vitro and in nude mice, these authors have shown that increased NO generation leads to altered cell migration. More importantly, it also leads to the development of resistance to taxol, a most commonly used anticancer drug for the treatment of breast cancer in the clinic. It is suggested that interferon (IFN-γ) may be responsible, in part, for these effects, as IFN-γ was found to induce NOS2, NO formation, and IL-6 formation in MDA-231 cells. IFN-γ has been linked to more aggressive, clinically resistant tumors [78,79].

One of the major problems in assessing the proof of mechanism in vivo is the complexity of living cells and humans. In assessing a mechanism of multi-drug resistance, it was possible to insert an mdr1 gene into tumor cells and assess the effects on cytotoxicity of various drugs of multi-drug resistance. However, no such experiments have been carried out with the iNOS gene. The situation is even more complex because the effective concentrations of NO required to induce resistance in tumor cells and in vivo is not clear now. In a series of publications, Radosevich et al. [80-82] have exposed human breast cancer cells to increasing concentrations of NO (via an NO-donor, DETA-NONOate) and examined the resulting biochemical and genetic changes associated with NO-adapted (resistant) cells. It was found that these NO-adapted cells grew faster and were significantly more aggressive. These cells expressed higher amounts of p53 protein and showed mutation of the p53 gene. Most interestingly, these cells showed up-regulation of GST-pi and were resistance to both X-ray and UV-radiation. Furthermore, these authors have reported that adapted cells also used more anaerobic glycolysis than the parent cells [82]. However, it is not known whether NO-adapted breast cancer cells maintained these biochemical changes when cultured under normal conditions in the absence of NO. In a similar sets of experiments, Radosevich et al. [83,84] have reported that when lung tumor (A549) cells were adapted to NO following exposure to increasing concentrations of DETA-NONOate they showed enhanced expression of key proteins involved in non-homologous and homologous recombination pathways for DNA repair. In additions, anti-apoptotic pathways were also enhanced as shown previously [76] following prolonged NO exposure.

Cheng et al. [85] have used various NO-donors (which mimic different intracellular exposures to NO) to examine gene expression profiles in order to identify genes involved in tumor response and resistance in human MDA-231 breast cells. Several genes were identified following NO exposure, including miR-21 (upregulated in various tumors and in cardiovascular diseases), heat shock protein-90, cystathionine-γ-lyase and IL-24. now, however, the roles of these genes in NO-induced resistance are not known.

Reversal of drug resistance by nitric oxide

Another interesting but conflicting effect of NO is that it also reverses mdr1-associated drug resistance. One of the hallmarks of mdr1 cells is that these cells overexpress p-170-glycoprotein (ppg) and multi-drug resistance proteins (MRP's), which remove intracellular drugs in an energy-dependent manner. Riganti et al. [86] have described the reversal of doxorubicin resistance when induction of NO production or delivery of NO via an NO-donor inhibited efflux of doxorubicin in HT-29-dox, a doxorubicin-selected resistant cells that overexpressed ppg. It was found that increased accumulation of doxorubicin promoted by NO was independent of the c-GMP pathway. It is interesting to note that NO also increased accumulation of doxorubicin in the sensitive HT-29 cells, suggesting that other mechanisms may also be involved in this. One of the mechanisms proposed to explain this reversal of resistance and increase in drug accumulation was that NO induced significant tyrosine nitration of the MRP3 transporter protein, which was also overexpressed along with ppg during the doxorubicin resistance selection process. Some tyrosine nitration was also detected in the sensitive HT-29 cells. Furthermore, it was suggested that S-nitration of cysteines may also be involved, leading to conformational changes and modification of activity of the transporter proteins. De Boo et al. [87] have confirmed that NO-induced nitration of tyrosine in the MRP3 protein resulted in the reversal of doxorubicin resistance in K562-dx cells which overexpressed both ppg and MRP3 proteins. It is interesting to note that in this dox-resistant cell line, nitro-ppg was also detected. Additionally, De Boo et al. [87] have shown that NO induced tyrosine nitration in ppg and MRP3 proteins in several other tumor cell lines expressing both ppg and MRP3 proteins, resulting in increased intracellular doxorubicin accumulation.

Curta et al. [88] have also shown that NO increased the accumulation of daunorubicin (an analog of doxorubicin) in leukemia K562 cells and enhanced apoptosis in the presence of daunorubicin. K562 cells constitutively express anti-apoptotic bcl2 and survivin proteins and are resistant to daunorubicin. Treatment with an NO-
donor in the presence of the drug decreased both bcl2 and survivin proteins, increased drug accumulation and enhanced the cytotoxicity of daunorubicin. K562 leukemia cells also overexpress both ABCC1 and lung-resistance proteins, which are a family of ABC transporters like pgp and MRP. While the role of lung-resistance protein overexpression in MDR is not clear, it has been shown to belong to the family of vault proteins and is involved in removing/redistributing drugs away from the nucleus [89,90], thus reducing effective drug concentrations in the nucleus. Tyrosine nitration of ABCC1 or lung-resistant protein, resistance-related proteins expressed in K562 cells, was not studied, even though both proteins were significantly decreased during the combined treatment with an NO-donor and daunorubicin. Ishima et al. [91] have reported that poly-S-nitrosated human albumin, which may serve as a reservoir for NO in vivo, increased doxorubicin intracellular concentration in a pgp-overexpressing K562-dx cells, resulting in the reversal of doxorubicin resistance in cells as well as in animals bearing K562-dx cells in vivo. It was found that poly-S-nitrosated albumin reversed doxorubicin resistance by decreasing the expression of pgp and HIF-1α in K562-dx cells.

It has been reported that NO formed either intracellularly or delivered via donor inhibits NF-kB activity and inactivates NF-kB-regulated downstream target anti-apoptotic genes [92]. Furthermore, this inhibition also results in the inhibition of the expressions of NF-kB-regulated transcription factor YY1, FAS and DR5 expressions, resulting in increased apoptosis and sensitization to various anticancer drugs. Huerta-Yepez et al. [93] have reported that high doses (0.5 mM to 1.0 mM) of DETA-NONOate sensitize resistant prostate cancer cells to cis-platin in vitro and in vivo. The NF-kB-regulated resistance factors YY1 and bclx were overexpressed in the resistant prostate cells, and it was shown that treatment with the NO-donor resulted in the inhibition of both YY1 and bclx. Combination treatment significantly enhanced cis-platin-induced apoptosis in vitro and decreased tumor growth in a xenograft model in vivo.

Studies indicate that activation of iNOS and production of NO in endothelial cells induces nitration of β-catenin and its translocation to the nucleus, causing activation of NF-kB and Wnt target genes [94]. Du et al. [95] have shown that increased Wnt/β-catenin signalling regulates INOS gene expression through NF-kB. An increase in Wnt/β-catenin signalling by iNOS results from the negative regulation of the Dickkopf-1 (DKK-1) gene. This interplay of Wnt signalling by iNOS is suggested to play an important role in drug resistance in non-small lung tumor cells by upregulating glutathione transferase-π and topoisomerase IIα but not pgp [96]. Insertion of DKK1 reversed this induction of GST-π and resistance to cis-platin [96].

Summary and Conclusions

Since the discovery of NO as the endothelial relaxing factor about 30 years ago, it has remained a most actively pursued area of research with over 100,000 publications. Currently, billions of dollars are being spent to bring NO-releasing agents to the treatment of various diseases. Many compounds, such as nitroglycerin, are available in the clinic for the treatment of acute angina. It is now very clear that NO is also involved in cancer progression, and resistance.

It appears that NO can be both good and bad in cancer chemotherapy. A multitude of studies and publications have now made a strong case that NO alone can induce cell death, and that it can be combined with several important clinical anticancer drugs to enhance/sensitize their activity against a variety of human malignancies. Efficacy and safety studies of GTN as a transdermal patch have been reported in a clinic trial against untreated stage IIIB/IV non-small cell lung cancer [97]. GTN was shown to have a positive response in combination with vinorelbine and cis-platin. Furthermore, a slow-release transdermal GTN patch has also been shown to be effective in men with prostate tumors whose disease was progressing following surgery or radiotherapy [98].

While several NO-donors have been considered and are being utilized in the clinic to treat heart-related complications [99], none of the NO-donors whether as a single agent or in combination appears to have entered into clinical trials or been successful beyond phase I study for cancer treatment. This is not very encouraging for those of us involved in basic and translational research with NO and NO-donors. This may, in part, be due to the fact that the biological effects of NO are multi-factorial without any clear-cut NO concentrations defined for the anti-cancer effects in vivo in humans. It is now well known that the actions of NO and NO-donors are dose-dependent, as at low concentrations NO promotes cancer cell survival, while at higher concentrations it induces DNA damage, apoptosis, and cell death. Furthermore, it is not clear how NO, can be cytotoxic to certain tumors in some cases, and in other cases, under similar conditions, it induces resistance to certain anticancer drugs by inducing bcl2 and HIF-1α stabilization in tumor cells.

While NO-donors have been invaluable in evaluating the chemistry and biological effects of NO, it is possible that some of the controversial reported biological effects of NO in cancer cell biology may be due to the NO-donors used. It is well known now that simple non-specific diolates, e.g. PPNO, DETACONOate, GSNO, spermine NONOate, and DEA-NONOate, release NO at physiological pH (and at different rates), but the biological effects of NO in tumor cells are diffusion controlled. Under this scenario, NO released outside the tumor cell must diffuse across the cell membrane to the site of action. We have found that higher concentrations of serum in the media affects the cytotoxicity of various NO-donors in tumor cells and requires higher concentrations of the NO-donors to achieve the same number of cells kill. While NO is not extremely reactive, it does react with a variety of substrates, including glutathione, protein -SH and heme-centers in proteins, resulting in a significant decrease in its effective concentrations and forming other products which may also have other deleterious biological effects. Under these conditions, a given biological response is not necessarily a true measure of the concentrations of NO-donor used. In contrast, studies carried out with those NO-donors that are target specific and require intracellular activation to release NO, e.g., JS-K, the effective concentration to achieve the same biological response would be different as there will be significantly fewer non-specific reactions with other biological molecules. Thus, it is necessary to exercise caution in understanding and interpreting the various biological effects described in the literature using different NO-donors.

It is very intriguing that NO can cause reversal of multidrug resistance by increasing intracellular drug concentrations in MDR/MRP-overexpressing cells. NO is reported to sensitize drugs to tumor cell killing under hypoxic environments. It is very important that we have a better understanding of the pathways leading to both sensitization to anticancer drugs in hypoxia and the reversal of drug resistance in cells overexpressing mdr1 and MRP proteins. Our preliminary work suggests that NO inhibits pgp functions by inhibiting ATPase, most likely via nitrosylation of the two -SH groups in ATPase. It is essential to find the right NO-delivering compound, which must also be very target specific to resistant tumor cells in the clinic. Such compounds would be ideal for the treatment of clinically resistant/or refractory tumors to successfully achieve our treatment goals.
Future Directions

Overcoming drug resistance in the clinic is a critical determinant for a successful therapy; one of the more promising directions at present is the use of NO-donors. The most successful course of action in developing cancer therapies with NO-donors will be to develop newer ones that are site-specific to tumor cells and release ‘NO following intracellular activation. Not only will this significantly reduce host toxicity, but it will also increase tumor cell killing. It is anticipated that such approaches will help to define the active ‘NO concentrations for both cytotoxicity and development of drug resistance. Tumor-specific intracellularly activated NO-donors would also be suitable for targeting ppG/MPR-overexpressing tumor cells for the reversal of drug resistance and/or increasing tumor sensitization to chemotherapy drugs in the clinic. Since ‘NO acts as a catalytic inhibitor of topoisomerase II, appropriately designed NO-donors should also be developed in the future for the treatment of tumors deficient in decatenation checkpoint.

Acknowledgements

We thank Dr. Ann Motten for help in editing the manuscript. We also thank Drs. Ron Mason, Maria Kadiiska and Thomas van’t Erve for their critical evaluation of the manuscript.

Funding

This research was supported [in part] by the intramural research program of the National Institute of Environmental Health Sciences, NIH. Statements contained herein do not necessarily represent the statements, opinions, or conclusions of NIHES, NIH, or the US Government.

References


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J Cancer Sci Ther, an open access journal
ISSN: 1948-5956


