Research Article Effect of TiO₂ Addition on the Properties of Al₂O₃-ZrO₂ Composites Prepared by Spark Plasma Sintering

I. Akin, E. Yilmaz, O. Ormanci, F. Sahin, O. Yucel, and G. Goller

Department of Metallurgical and Materials Engineering, Istanbul Technical University, Istanbul 34469, Turkey Address correspondence to G. Goller, goller@itu.edu.tr

Received 14 January 2010; Accepted 3 February 2010

Abstract In this study, monolithic Al₂O₃ and Al₂O₃/3Y-TZP (90–10 vol%) and Al₂O₃/3Y-TZP with 5 wt% TiO₂ composites were prepared by spark plasma sintering (SPS) at temperatures of 1350, 1460 °C and 1300 °C for 300 s under a pressure of 40 MPa, respectively. Shrinkage of the specimens during SPS process was continuously monitored. Densities of the composites were determined by the Archimedes' method. Vickers hardness (H_V) was measured under loads of 9.8 N. Fully dense Al₂O₃ and Al₂O₃/3Y-TZP composites containing 0 and 5 wt% TiO₂ with a relative density of 99% were obtained. Vickers hardness of the composites decreased with increasing TiO₂ content from 19.8 ± 0.4 GPa to 17.3 ± 0.3 GPa.

Keywords spark plasma; sintering; alumina; zirconia; titania

1 Introduction

Monolithic alumina and zirconia ceramics are widely used for orthopedic implants such as total hip and knee replacement prostheses due to their excellent mechanical properties, high wear resistance, and biocompatibility. It is well known that yttria stabilized tetragonal zirconia (3Y-TZP) has higher strength and fracture toughness than alumina, because of transformation toughening mechanism. A new generation ceramic for joint prostheses, zirconia toughened alumina composites, possesses higher fracture strength and toughness compared to monolithic alumina [2,4,5].

The effects of additives for alumina ceramics have been aimed to decrease the sintering temperature, improve microstructure and properties. The effect of TiO₂ addition has been reported to promote the sintering behavior and grain growth phenomena in several studies [5]. TiO₂ addition enhanced diffusivity due to the Al³⁺ vacancies, as generated by Ti⁴⁺ substituting for Al³⁺ [1,3,5].

The purpose of this study is to investigate the sintering behavior, densification and hardness of the Al_2O_3 , $Al_2O_3/3Y$ -TZP (90–10 vol%) and $Al_2O_3/3Y$ -TZP (90– 10 vol%) with 5 wt% TiO₂ composites prepared by spark plasma sintering (SPS) method. The phase analysis and microstructures of the specimens were also investigated.

2 Materials and methods

Al₂O₃ (Baikowski Grade SM8, France, an average particle size of $0.6 \,\mu\text{m}$), 3Y-TZP (Tosoh Grade TZ-3Y, Japan, an average particle size of 0.05–0.1 μ m) and TiO₂ (Anatase, Merck, product code 1.00808, Germany) powders were used as starting materials (Table 1). The raw materials were weighed in appropriate quantities, ball milled in ethanol for 24 h and then dried. A graphite die 50 mm in inner diameter was filled with the mixture, followed by sintering using an SPS apparatus (SPS-7.40 MK-VII, SPS Syntex Inc.). Pure Al₂O₃ was sintered at 1350 °C, Al₂O₃/3Y-TZP composites with and without TiO2 were sintered at 1300 and 1460 °C with a heating rate of 1.7 °C/s in vacuum. A uniaxial pressure of 40 MPa and pulsed direct current (12 ms/on, 2 ms/off) were applied during the entire process. The current was controlled manually during monitoring the displacement behavior of the samples. The crystalline phases were identified by X-ray diffractometry (XRD; Rigaku MiniFlex) in the 2θ range of $10-80^{\circ}$ with Cu K α radiation. The densities of the specimen were determined by Archimedes' method and converted to relative density using theoretical densities of Al2O3, yttria stabilized ZrO_2 and TiO_2 . The fracture surface of the samples were coated with a thin film platinum and subjected to microscopic investigation by a field emission scanning electron microscope (FE-SEM; JEOL JSM-7000F). Microhardness tests were applied to the polished samples under a constant load of 9.8 N with 12 s indentation time.

Material grade	Supplier	$Y_2O_3 \;(mol\%)$	$Al_2O_3 (wt\%)$	Grain size ^a (nm)
TZ-3Y	Tosoh	3	0.1	27

^a According to the supplier datasheets.

Table 1: ZrO₂ starting powders.



Figure 1: Displacements of the composites, and the time dependence of isothermal displacement at sintering temperatures.

3 Results and discussion

The densification of the specimens during SPS process was evaluated by the displacement of punch rods due to the shrinkage of the samples. Figure 1 shows the displacements of Al_2O_3 and $Al_2O_3/3YTZP$ composites without TiO₂ and with 5 wt% TiO₂, and isothermal shrinkage at sintering temperatures.



Figure 2: XRD patterns of Al_2O_3 and $Al_2O_3/3Y$ -TZP composites.

The shrinkage of Al_2O_3 started at 950 °C and completed at 1310 °C. The starting temperature of shrinkage (1060 °C) for the $Al_2O_3/3$ Y-TZP composites containing 5 wt% TiO₂ was significantly lower than that of $Al_2O_3/3$ Y-TZP (1200 °C). Thus, the presence of 5 wt% TiO₂ promoted the densification of $Al_2O_3/3$ Y-TZP and decreased the sintering temperature of $Al_2O_3/3$ Y-TZP composites from 1460 to 1300 °C.

The XRD patterns of the Al₂O₃ and Al₂O₃/3Y-TZP composites sintered at different temperatures for 300 s are shown in Figure 2. X-ray analysis indicated that α -Al₂O₃ and tetragonal ZrO₂ became fully crystalline and monoclinic ZrO₂ peaks did not appear at the end of the sintering process.

Figure 3 shows the Vickers hardness of Al_2O_3 and $Al_2O_3/3Y$ -TZP composites at load of 9.8 N. $Al_2O_3/3Y$ -TZP (90–10 vol%) composite had higher hardness than



Figure 3: Vickers hardness of Al₂O₃ and Al₂O₃/3Y-TZP composites.



Figure 4: Fracture surface micrographs of (a) monolithic Al_2O_3 sintered at 1350 °C, (b) Al_2O_3 -3Y-TZP sintered at 1460 °C, and (c) Al_2O_3 -3Y-TZP + 5TiO₂ sintered at 1300 °C for 300 s.

the monolithic Al_2O_3 . The hardness of Al_2O_3 slightly increased with the addition of 10 vol% yttria stabilized (3 mol%) ZrO₂ (3Y-TZP) from 19.8 to 20.2 GPa. The addition of 5 wt% TiO₂ decreased the Vickers hardness of $Al_2O_3/3$ Y-TZP composite from 20.2 to 17.3 GPa. This relatively low hardness could be mostly due to the 2 wt% porosity.

Microstructures of fracture surfaces of Al_2O_3 and $Al_2O_3/3Y$ -TZP composites are shown in Figure 4. In Figure 4(a), Al_2O_3 consisted of both large and small

equiaxed grains $0.5-3 \,\mu\text{m}$ in size and straight grain boundaries. Figure 4(b) demonstrates the SEM image of fracture surface of Al₂O₃-3Y-TZP. In this figure, alumina grains (grey) in size of $0.6-1.5 \,\mu\text{m}$ and 3Y-TZP grains (white) in size of $0.05-0.1 \,\mu\text{m}$ are shown. Figure 4(c) is the SEM micrograph of Al₂O₃/3Y-TZP composites containing 5 wt% TiO₂. The grain size of alumina was not significantly changed by the addition of TiO₂.

4 Conclusions

 Al_2O_3 and $Al_2O_3/3Y$ -TZP composites were prepared by SPS at temperatures of 1300–1460 °C for 300 s under 40 MPa. The addition of 5 wt% TiO₂ improved densification of $Al_2O_3/3Y$ -TZP and decreased the sintering temperature of $Al_2O_3/3Y$ -TZP composites from 1460 to 1300 °C. The presence of 10 vol% 3Y/TZP slightly increased the hardness of Al_2O_3 and suppressed the grain growth of alumina.

Acknowledgment The authors would like to thank H. Dincer due to his contribution for SPS processes and T. T. Alpak for electron microscopy investigations.

References

- R. D. Bagley, I. B. Cutler, and D. L. Johnson, *Effect of TiO₂ on initial sintering of Al₂O₃*, J Am Ceram Soc, 53 (1970), pp. 136–141.
- [2] J. Chevalier and L. Gremillard, *Ceramics for medical applications: a picture for the next 20 years*, J Eur Ceram Soc, 29 (2009), pp. 1245–1255.
- [3] K. Hamano, C.-S. Hwang, Z. Nakagawa, and Y. Ohya, *Effects of TiO₂ on sintering of alumina ceramics*, J Ceram Assoc, 94 (1986), pp. 505–511.
- [4] C. Santos, M. H. Koizumi, J. K. M. F. Daguano, F. A. Santos, C. N. Elias, and A. S. Ramos, *Properties of Y-TZP/Al₂O₃ ceramic nanocomposites obtained by high-energy ball milling*, Mater Sci Eng A, 502 (2009), pp. 6–12.
- [5] C.-J. Wang and C.-Y. Huang, Effect of TiO₂ addition on the sintering behavior, hardness and fracture toughness of an ultrafine alumina, Mater Sci Eng A, 492 (2008), pp. 306–310.