

Cytotoxicity and Reactive Oxygen Species Generated by Ferrocenium and Ferrocene on MCF7 and MCF10A Cell Lines

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Abstract

The aim of this study was to investigate the effect of ferrocene (FeCp_2) and ferrocenium salt (FeCp_2BF_4) on the viability of MCF7 breast cancer and MCF10A non-tumorigenic epithelial cells and the role of Reactive Oxygen Species (ROS) production in cell cytotoxicity. FeCp_2BF_4 displayed higher cytotoxicity than FeCp_2 , and the cell type contributes toward complex toxicity, as MCF7 cells displays greater toxicity than MCF10A cells. The mechanism of toxicity seems to involve the generation of ROS, with MCF7 cells producing higher levels than epithelial cells. In addition, the inhibition of ROS was found to be protective against ferrocene induced cell death. The findings of cancerous cell-induced cytotoxicity by ROS indicate a potential utility of ferrocenyl derivatives in the treatment of cancer.

Keywords: Ferrocene; Ferrocenium; Reactive oxygen species; MCF7 cells; MCF10A cells

Introduction

The accidental discovery of bis(cyclopentadienyl) iron (II) (ferrocene) in 1951 has played an important role in the development of organometallic chemistry [1,2]. Ferrocene (FeCp_2) has been well studied due to its properties and applications in research areas ranging from organic synthesis, catalysis and material science [3-6] (Figure 1). The redox properties as well as its thermodynamic and kinetic stabilities have made ferrocene one of the most widely studied organometallic complexes. The neutral, uncharged complex can undergo one-electron oxidation producing ferrocenium cation. Ferrocenium (FeCp_2BF_4) has an unpaired electron in one of the non-bonding orbitals (e_g) and as a result it is a free radical species of high stability. With these chemical properties, it is not surprising that the $\text{FeCp}_2/\text{FeCp}_2\text{BF}_4$ system has found fertile grounds in the biomedical field.

In 1984 Köpf-Maier and co-workers [7] reported the anticancer properties of FeCp_2BF_4 salts in an Ehrlich ascite tumor. Ferrocenium (induce DNA oxidative damage and 8-oxoguanine as a initial product of guanine oxidation [8,9]. However, if FeCp_2 is transported into the tumor cells, once uptaken, it generates H_2O_2 by autooxidation, forming Fc^+ [10]. To improve FeCp_2 uptake by cells, these ferrocenyls have been functionalized with nucleic acids, proteins, hormones, selective endocrine modulators and carbohydrates [11-13].

In the past years, our group has initiated a project investigating the antitumor properties of modified ferrocene (ferrocenyls), in particular modifying the structure with pendant groups on the Cp ring to make more active and selective anticancer agents. But, there are mechanistic

details on FeCp_2 that need to be investigated in more depth. In this regard, the mechanistic studies reported independently by Osella and Kovjazin et al. [9,10] might be contradictory. For instance, Osella and collaborators [9] demonstrated that the cytotoxic effect of ferrocenium is associated to the production hydroxyl radicals under physiological conditions and nuclear DNA, cell membrane and topoisomerase II are the possible targets, whereas ferrocene has *no inhibitory* effect on cancer cells [8,9]. In contrast, Kovjazin and collaborators [10] reported that ferrocene can produce free radical species and its antitumor activity is mediated by immune stimulation. Thus, we decided to investigate the ROS production of ferrocenium versus the non toxic FeCp_2 and correlate it with their cytotoxicity in the MCF7 breast cancer cell line. Herein we report our findings.

Materials and Methods

Cell culture

Human breast adenocarcinoma MCF7 cells (ATCC, Cat. HTB-22) were all adapted and maintained in MEME supplemented with 2 mM L-glutamine, 1 mM sodium pyruvate, 0.01 mg/ml bovine insulin, 10% fetal bovine serum (FBS) (all reagents from Sigma-Aldrich) and 100 units/ml penicillin G/100 $\mu\text{g}/\text{ml}$ streptomycin sulfate (Lonza, Cat. 17-603E). Human mammary epithelial MCF10A cells (ATCC, Cat. CRL-10317) were maintained in MEME (MEGM bullet kit, Lonza, Cat. CC-3150) supplemented with 13 mg/ml BPE, 0.5 mg/ml hydrocortisone, 10 $\mu\text{g}/\text{ml}$ hEGF, 5 mg/ml bovine insulin, 100 ng/ml cholera toxin and 100 units/ml penicillin G/100 $\mu\text{g}/\text{ml}$ streptomycin sulfate (Lonza, Cat. 17-603E). Cell cultures were incubated in a humidified 5% CO_2

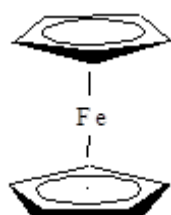


Figure 1: Structure of ferrocene.

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atmosphere at 37°C. Strict attention was paid to using cells when in the logarithmic phase of cell growth. Cells were seeded in 96-well plates at a final concentration of 5×10^4 cells/ml.

Cell viability

Cell viability was assessed using the cell permeant and non-fluorescent probe calcein AM (Anaspec, Cat. 89201). Cells were seeded in 96-well plates at a concentration of 5×10^4 cells/ml and incubated overnight. Cells were then washed with Phosphate Buffered Saline solution (PBS) and incubated with 4 μ M calcein AM, during 40 min at 37°C and washed again with PBS. Antibiotic-free medium containing various concentrations of the ferrocene complexes in 5% DMSO/95% medium were added to the wells. The concentrations used for FeCp₂ were 25, 50, 100, 250, 500, 1000, 1500, 2000, 2500 and 3000 μ M and for FeCp₂BF₄ 2.5, 5, 10, 25, 50, 100, 150, 200, 250 and 300 μ M. Three independent measurements were carried out for every concentration. After 1, 2, 3, 4, 12, 24 and 48 h of FeCp₂ or FeCp₂BF₄ exposure, the fluorescence was measured at 530 nm. Three controls were set up for each experiment: 1) IC₀ consisting of cells without the ferrocene complexes (representing 100% growth), 2) IC₁₀₀ consisting of medium alone (representing 0% growth) and 3) cells growing on 5% DMSO/95% medium (to determine the cytotoxic effect of DMSO in the medium). Background fluorescence, due to the non-specific binding was subtracted from the measurements of exposed cells. For IC₅₀ (50% inhibitory concentration) calculations, survival data were analyzed by dose-response curve fitting using Prism 4.0 (GraphPad, San Diego, CA). Fluorescence was measured using a Perkin Elmer HTS 7000 (MA, USA) microplate reader.

Measurement of reactive oxygen species (ROS)

To assess ROS production, MCF7 and MCF10A cells were treated with the oxidation-sensitive dye, 2',7'-dichlorodihydrofluorescein diacetate (H₂DCF-DA, Invitrogen, Cat. D399). The oxidation product of H₂DCF-DA generates the fluorescent 2',7'-dichlorofluorescein (DCF) which has excitation/emission maxima of ~495/529 nm. Cells were seeded in 96-well plates at a concentration of 5×10^4 cells/ml and cultured with different concentrations of FeCp₂ or FeCp₂BF₄ (as described above). After 1, 2, 3, 4, 12, 24 and 48 h of treatment, cultures were loaded with 10 μ M of H₂DCF-DA and incubated for 40 minutes at 37°C, after which time ROS production was evaluated at 530 nm, with H₂O₂ as positive control.

ROS quenching

To determine ROS production in FeCp₂- or FeCp₂BF₄-induced cell death, MCF7 and MCF10A cells were seeded in 96-well plates at a concentration of 5×10^4 cells/ml. N-acetyl cysteine (NAC, Sigma-Aldrich Cat. A9165) was added to the cells at 10 mM for 1 h. After NAC pretreatment, cells were cultured with FeCp₂ or FeCp₂BF₄ (at the concentrations described above). ROS production was determined by adding H₂DCFDA to pretreated cell cultures and recording the DCF fluorescence at 530 nm.

Statistical analysis

All data was analyzed using GraphPad Prism 4 software (GraphPad Software, Inc., San Diego, CA). Data for the viability assay was transformed, normalized and statistically analyzed using a dose-response curve fitting with significance levels defined as $p < 0.05$. Statistical analysis for the ROS production data was performed using a two-way ANOVA.

Results

Morphological evaluation of MCF7 and MCF10A Cells by microscopy

The morphological changes on MCF7 cells were visualized by inverted microscopy, after exposing the cells to different concentrations of ferrocenes (as described previously). The MCF7 cells without treatment (used as control) show a uniform distribution maintaining an integral cellular membrane. Given the fact that FeCp₂ is not toxic at low concentrations and ferrocenium cytotoxicity is low, we used high concentrations for the study to detect the ROS and morphological changes. After 12 h of exposure, MCF7 cells treated with up to 500 μ M FeCp₂ showed no significant difference when compared with controls (data not shown). However, at concentrations of 1000 μ M FeCp₂ or higher, MCF7 cells were detached and morphologically rounded, different when compared with the controls. Similar morphological changes were observed in MCF10A cells (data not shown). Morphological evaluation by microscopy was also performed for MCF7 and MCF10A cells at longer drug exposure times. At 24 h exposure, with 1000, 1500, 2000 and 3000 μ M of FeCp₂, MCF7 cells exhibited morphological changes characteristic of apoptosis, including karyopycnosis, oversized cytoplasmic particles, damaged organelles and rupture of the cytoplasmic membrane and nuclear envelope. Figure 2A shows the morphological changes in MCF7 cells after exposure to maximum concentration of FeCp₂ during 24 h. More significant morphological changes were observed after 48 h of exposure time. For MCF10A cells, similar morphological changes to those of MCF7 were obtained at 24 and 48 h (data not shown).

Morphological changes induced by FeCp₂BF₄ on MCF7 and MCF10A cells were also examined. After 12 h of exposure, MCF7 and MCF10A cells to FeCp₂BF₄ at a concentration below 50 μ M, these showed no significant difference when compared with controls. However, at concentration between 100 and 300 μ M of FeCp₂BF₄, MCF7 cells were detached and morphologically rounded. Similar morphological changes were seen with MCF10A cells upon exposure to 100-300 μ M FeCp₂BF₄. After 24 h of exposure to 100-300 μ M FeCp₂BF₄, MCF7 cells exhibited almost identical morphological changes as those observed when exposed to FeCp₂ but at concentrations an order of magnitude higher. These include apoptosis with karyopycnosis, oversized cytoplasmic particles, damaged organelles and ruptured cytoplasmic membrane and nuclear envelope. Figure 2B shows the morphological changes in MCF7 cells after a 24 h exposure to 300 μ M FeCp₂BF₄. These results indicate that ferrocenium is more active than FeCp₂ in inducing cell damage. At 50 μ M there was a reduction in the number of the cells and the remaining cells seemed attached but

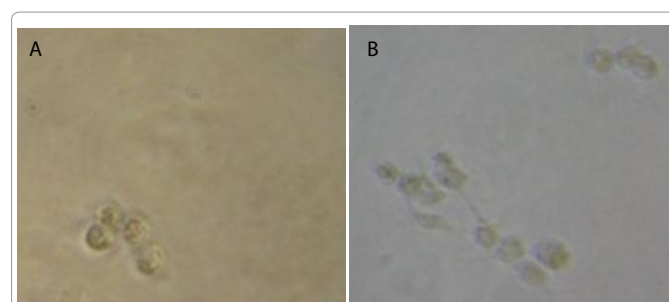
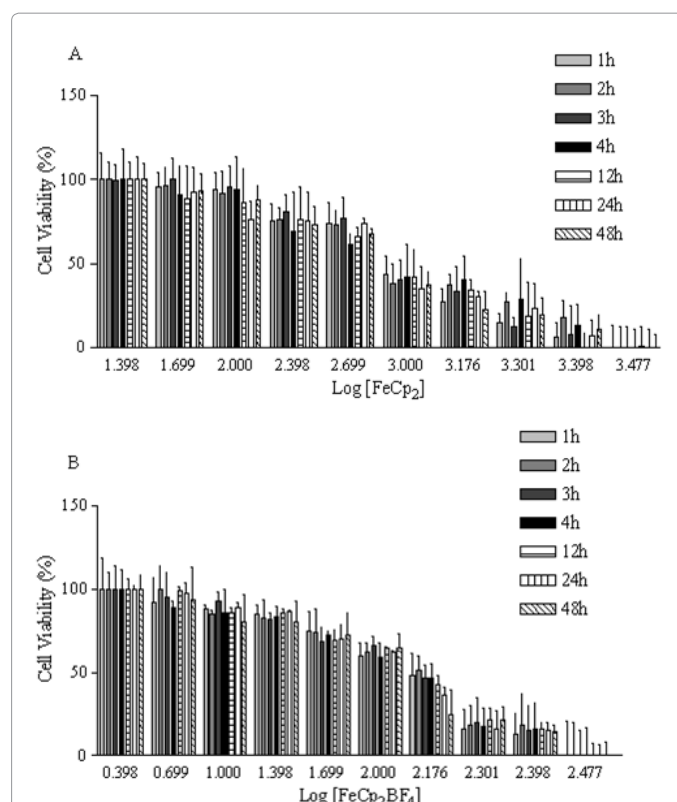
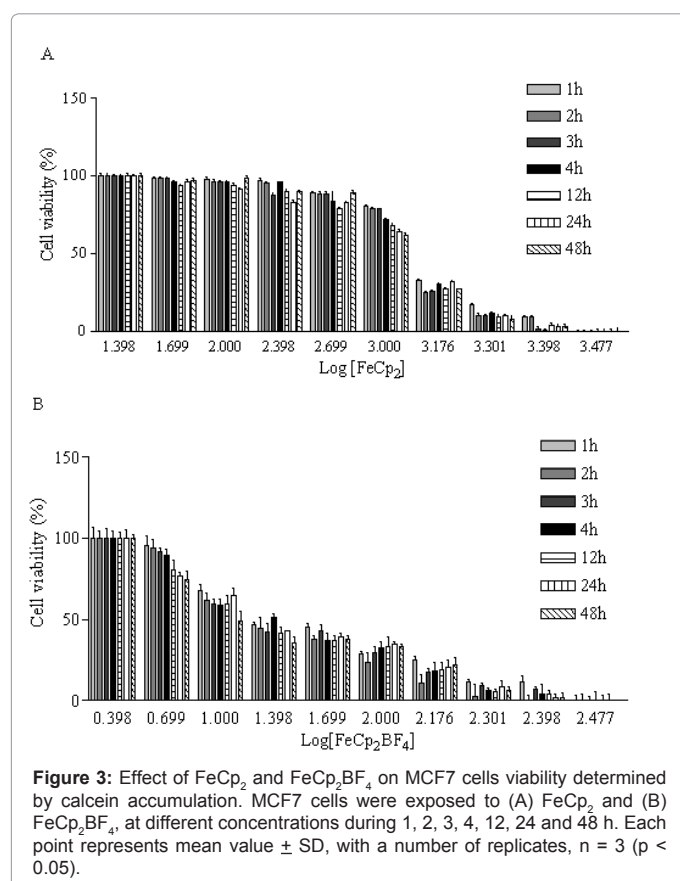


Figure 2: MCF-7 cells exposure to (A) FeCp₂ 3000 μ M or (B) FeCp₂BF₄ 300 μ M during 24h. Cells were visualized by inverted microscope. Magnification, 10X, insets, 40X.

morphologically different (rounded) when compared with controls. A similar effect was observed on MCF10A cells (data not shown).

Cell viability studies

To determine if FeCp_2 and FeCp_2BF_4 toxicity was dependent upon the cellular microenvironment, the cytotoxic activity of these compounds was measured on MCF7 cells and compared to MCF10A cells. Both types of cells were treated with 25-3000 μM FeCp_2 or with 2.5-300 μM FeCp_2BF_4 , at time intervals of 1, 2, 3, 4, 12, 24 and 48 h. Figure 3 shows the viability of MCF7 cells upon exposure to FeCp_2 and FeCp_2BF_4 as function of time. Curve fitting of the survival data of the cell lines in response to various concentrations of the agents was used to calculate IC_{50} values. For MCF7 cells, FeCp_2 displayed an IC_{50} of 1421 ± 5.3 μM (Figure 3A) and 261.98 ± 1.5 (Figure 4B) were determined at 24 h. These results are in agreement with previous published data [11], which show a significant dose-dependent antiproliferative effect of these complexes. These IC_{50} values correspond very well with the concentrations of the complexes that induced significant morphological changes. Evaluation of the MCF7 and MCF10A cells showed that after 12 h at concentrations above 1000 μM for FeCp_2 or above 100 μM for FeCp_2BF_4 , these compounds induced morphological changes in cells, with FeCp_2BF_4 showing a higher cytotoxicity than FeCp_2 . For MCF7 cells it can be seen that at log values of about 3.000 and 0.699 for FeCp_2 and FeCp_2BF_4 , respectively, there is a marked drop in cell viability which increases with time. At higher concentrations the drop in cell viability is less pronounced and is less dependent on time. In the case of MCF10A this effect is more subtle and the drop in cell viability decrease steadily with concentration. The stronger effect on



cell viability of FeCp_2BF_4 over FeCp_2 seen with MCF7 can also be seen in MCF10A cell, with similar IC_{50} values for both cell lines. 1.5 μM for FeCp_2BF_4 (Figure 3B), whereas for MCF10A cells an IC_{50} of 1250 ± 8.3 μM for FeCp_2 (Figure 4A) and 126 ± 13.79 μM for FeCp_2BF_4 .

Measurement of ROS production induced-ferrocenes in MCF7 and MCF10A cells

Experiments were performed to determine whether the MCF7 and MCF10A cytotoxicity from FeCp_2 or FeCp_2BF_4 exposure could be related to the generation of intracellular ROS. Cellular production of ROS was measured at 1, 2, 3, 4, 12, 24 and 48 h after FeCp_2 or FeCp_2BF_4 exposure, using the $\text{H}_2\text{DCF-DA}$ dye. There was a significant ($p < 0.05$) increase in ROS in MCF7 cells at FeCp_2 concentrations above 1000 μM , as shown in Figure 5A (Supplementary Figure S1). This concentration is a threshold before which exposure time played a significant role in ROS production. In addition, this threshold concentration correlates with the point at which there was a significant drop in cell viability. Results for MCF10A cells show a similar threshold concentration at which there is a significant increase in ROS and that this correlates with a drop in cell viability, although at lower levels than for MCF7 cells (Figure 6A). Higher levels of ROS were detected when exposing MCF7 and MCF10A cells to FeCp_2BF_4 (Figures 5B and 6B; and Supplementary Figure S2), although two fold lower levels were detected for the latter. Previous studies have reported that the antitumor properties shown by exposure to FeCp_2 could be due to the metabolic formation of ferrocenium ions (Cp_2Fe^+), which in turn induce oxidative damage

to DNA [12,13]. A possible mechanism for the action of ferrocenyl derivatives have been proposed in T lymphocytes, in which, p21^{ras} activates T-cells through an oxidative stress-mediated mechanism that elicits an immune stimulatory effect [10].

Role of ROS in FeCp₂ or FeCp₂BF₄ induced cytotoxicity

To provide further evidence that ROS production is in fact being generated due to FeCp₂ or FeCp₂BF₄ exposure, the antioxidant NAC was added prior to treatment with FeCp₂ or FeCp₂BF₄ to quench ROS. Supplementary Figures S1A and S1B, shows that NAC has significant effect in preventing ROS induced by FeCp₂ or FeCp₂BF₄ on MCF7 and on MCF10A cells (Supplementary Figure S3 and S4). In this study significant differences were observed between treated with 10 mM NAC, for each concentration tested of FeCp₂ and FeCp₂BF₄. These results are in agreement with previous studies showing the crucial role of ROS in nanoparticle (NP)-induced cytotoxicity, in which cells exposed to 10 mM NAC cells remain viable even at NP concentration previously shown to reduce cell viability below 10% [14].

Discussion

In this study we correlate the cytotoxic effect of FeCp₂ and FeCp₂BF₄ on MCF7 breast cancer cells and MCF10A mammary epithelial cells with the ROS production. Here we present novel findings demonstrating that cancerous breast cells are markedly more susceptible (i.e., 2-3 times) to FeCp₂ or FeCp₂BF₄-mediated toxicity than their normal counterparts. In addition, cell show an order of magnitude lower IC₅₀ values for FeCp₂BF₄. The higher cytotoxicity shown by FeCp₂BF₄ over FeCp₂ could be explained mainly by the former agent being a water-

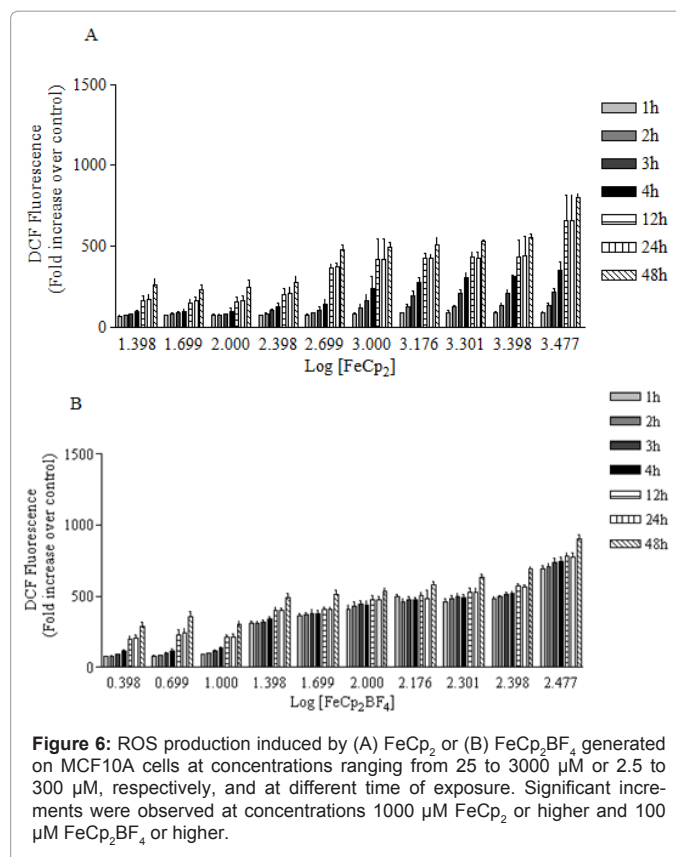


Figure 6: ROS production induced by (A) FeCp₂ or (B) FeCp₂BF₄ generated on MCF10A cells at concentrations ranging from 25 to 3000 μM or 2.5 to 300 μM, respectively, and at different time of exposure. Significant increments were observed at concentrations 1000 μM FeCp₂ or higher and 100 μM FeCp₂BF₄ or higher.

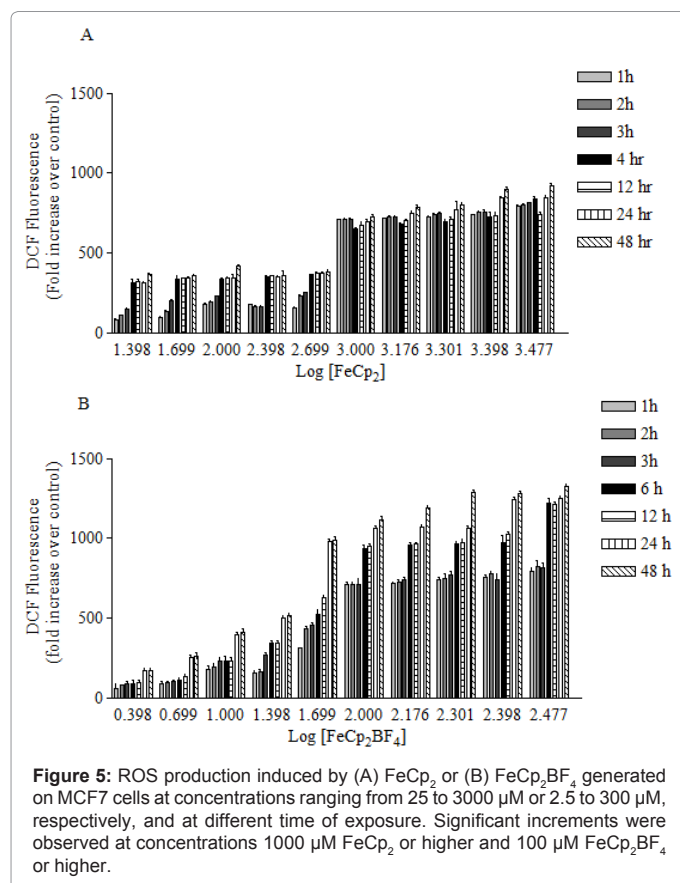


Figure 5: ROS production induced by (A) FeCp₂ or (B) FeCp₂BF₄ generated on MCF7 cells at concentrations ranging from 25 to 3000 μM or 2.5 to 300 μM, respectively, and at different time of exposure. Significant increments were observed at concentrations 1000 μM FeCp₂ or higher and 100 μM FeCp₂BF₄ or higher.

soluble stable cation, which has better access to the negatively-charged MCF7 and MCF10A surface [15]. These findings may be of important clinical interest as one of the greatest challenges facing chemotherapy is the inability of anticancer drugs to effectively distinguish between normal and cancerous tissues. These findings are in agreement with previous studies, showing that rapidly dividing tumor cells are more susceptible to different activation stimuli resulting in different toxicity responses, when compared to quiescent cells. The preferential killing of the rapidly dividing cancer cells relative to quiescent cells of the same lineage suggest that the mechanisms of these complexes toxicity might be related to the proliferative potential of the cell [14]. Our findings that the cell viability was related to the ROS production, suggest that increasing in ROS levels exceeds the capacity of the cellular antioxidant defense system, and causes cells to enter a state of oxidative stress which results in damage of cellular components such as lipids, protein and DNA [16]. The oxidation of fatty acids then leads to the generation of lipid peroxides that initiate a chain reaction leading to disruption of plasma and organelle membranes and subsequent cell death. The persistent oxidative stress that characterizes cancer cells has been proposed as the basis for a strategy to develop new therapeutic strategies based on ROS-inducing agents [17]. Based on our results, ferrocenes and their derivatives seem as good candidates for this therapeutic strategy. In addition, our findings reveals a concentration and time dependent increase in ROS production in both cell types, with higher levels been observed in MCF7 cancerous cells consistent with the ability of these cells to generate large amounts of ROS. In fact, our studies using the ROS quencher, NAC, demonstrated the causal role of ROS generation in FeCp₂ or FeCp₂BF₄-mediated cytotoxicity. These findings further support that functionalization of ferrocene with

pendant groups that facilitate the entrance of ferrocene inside the cell and modulate the redox chemistry inside the cell is worth pursuing [11-13,18-23]. For instance, a series ferrocenyls functionalized with selective endocrine receptor modulator (ERM) have reported [19]. These ferrocifens elicit their antiproliferative effects mediated by the pendant groups (ERM) which act as a antagonist and a redox antenna [19]. Therefore, lipophilicity, molecular recognition and redox chemistry must be considered when designing ferrocenyl anticancer agent [8-23].

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