



## Biomimetic Synthesis and Characterization of Silver Nanoparticles Synthesized by Leaf Extract of *Bridellia stipularis* L Blume and Evaluation of their Effect on Mitotic Chromosomes of *Pisum Sativum* L

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### Abstract

The present investigation focuses on the synthesis of silver nanoparticles using leaf extract of *Bridellia stipularis* and their effect on mitotic chromosomes of root meristem of *Pisum sativum*. The phytochemical silver nanoparticles were monitored by UV-Vis spectroscopy, FTIR, XRD, AFM and HR-TEM analysis. The ultraviolet-visible spectrum shows a prominent peak at  $\lambda_{max}$  = 413 nm at pH 10. The size and shape of the AgNPs were found to be 5 to 85 nm and are spherical in nature. This study focuses on assessment of toxic effect of AgNPs on mitotic chromosomes of root meristematic cells of *P. sativum* (2n=14). *P. sativum* root cells were treated with four different concentrations 4, 8, 12, 16  $\mu$ g/ml of the AgNPs solution at the interval of 6, 12, 18 and 24 hours duration. The mitotic index was decreased ( $55.43 \pm 5.69\%$  to  $13.99 \pm 0.60\%$ ) with an increase in the number of various chromosomal aberrations ( $36.95 \pm 3.15\%$ ) at higher concentration. It is found that these chromosomal aberrations include chromatid bridge, sticky, diagonal chromosome, c-metaphase, multibrIDGE and micronucleus. The results revealed that the percentage of mitotic index (MI) is inversely proportional and chromosomal aberrations (CAs) are directly proportional to the concentration and duration of exposure. The results were found to be statistically significant at  $p < 0.05$ . It is evident from the results that AgNPs cause significant inhibition of mitotic index and increased chromosomal abnormalities.

**Keywords:** *Bridellia stipularis*; Silver nanoparticles; Cytotoxicity; Genotoxicity; *Pisum sativum* assay

### Introduction

Over several decades, the noble metal nanoparticles have gained a lot of public interest due to increased application of nanomaterials within a size range of 1-100 nm in many areas of human endeavors [1]. Nowadays, nanotechnology is considered as the technology of 21<sup>st</sup> century, and it can be considered as an important bridge between nanotechnology and all available kinds of renewable energies [2-6]. The production of functional nanoparticles (gold, silver, zinc etc.) is versatile in the field of green nanotechnology [7]. Among them, silver nanoparticles are most widely used as anticancer, antioxidant, antifungal, antidiabetic, antiparasitic and antituberculosis agent [8-17]. Phytochemicals present in the leaf extract may act as both reducing and stabilizing agents for the synthesis of silver nanoparticles. These nanoparticles possess unique physical, chemical, electrical, magnetic, mechanical and biological properties [18]. Currently, metal nanoparticles are synthesized by various methods in high technological areas such as chemical [19-21], electrochemical [22], Magnetic and thermal properties [23], radiation [24], photochemical [25], and biological methods [26]. It was proved by many researchers that the exploiting nanofluid in solar systems, offers distinctive advantages over conventional fluids [27-29]. Nanofluids enhance the rate of heat transfer in many thermal generation industries. The physical and chemical methods have certain disadvantages due to involvement of hazardous by-products during their synthesis of nanomaterials and these chemicals pose a serious threat to the environment and human health. While, biological methods are reliable, cost-effective, environmentally benign, suitable for large scale production and these methods are more advantageous over the physicochemical methods [30-33]. Many investigators synthesized nanoparticles by using microorganisms [34], enzymes [35], fungi [36-38], and higher plants [39-42].

The *Bridellia stipularis* (L.) Blume plant species is used as a remedy for several diseases by traditional herbal healers for the treatment of jaundice, allergy, inflammation, scabies and anemia [43-46]. Because these plant species contain various medicinally important secondary

metabolites and used for the production of anthocyanin pigments in plants [47]. In the present investigation, silver nanoparticles were synthesized by leaf extract of *B. stipularis* and used for the evaluation of their cytotoxicity on mitotic chromosomes of root meristems of *P. sativum* which serves as a genetic model system to study genetic endpoints like mitotic index (MI), chromosomal aberrations (CAs), nuclear abnormalities (NAs) and micronucleus (MN) in plant system [48]. The literature review reveals that there are no reports on cytotoxicity of silver nanoparticles on mitotic chromosomes of *P. sativum*.

### Materials and methods

#### Plant material collection

The leaves of *B. stipularis* (synonym *Bridellia scandens*) belonging to the family Euphorbiaceae were collected from the Botanical Garden of Karnatak University campus, Dharwad, Karnataka India.

#### Preparation of aqueous leaf extract

The collected leaves of *B. Stipularis* were cleaned with running tap water to remove adhered dust impurities and other contaminants, followed by double-distilled de-ionized water and shade dried at room temperature. About 10 g of shade dried leaves were chopped into small pieces and transferred to 250 ml Erlenmeyer flask containing 100 ml of de-ionized water. Further, it was heated for 15-20 minutes

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at 60 °C, until the color of the reaction mixture changes from watery to light yellow color. The aqueous extract was allowed to cool at room temperature and then filtered through Whatman no. 1 filter paper. The reaction mixture was stored in a refrigerator at 4 °C for further analysis.

### Phytosynthesis of silver nanoparticles

Synthesis of AgNPs has been carried out by adding 5 ml leaf extract of *B. stipularis* to 250 ml of the Erlenmeyer flask containing 95 ml of 1 mM silver nitrate ( $\text{AgNO}_3$ ) solution. Reduction of silver ions to silver nanoparticles was observed visually within the next 30 minutes by the appearance of light yellowish to dark brown color of the reaction mixture indicates the formation of AgNPs in the solution (Figure 1).

### Characterization of phyto-genic AgNPs

The change in color of the reaction mixture from yellow to dark brown color indicates the bioreduction of silver ions to silver nanoparticles. Formation of AgNPs was confirmed by characteristic absorption spectrum (JascoV-model 670, UV-Vis NIR spectrophotometer) with a resolution of 1 nm, operated at a wavelength of 300 to 700 nm. The nanoparticles solution was centrifuged for 30 minutes at 35000 rpm (Remi R-8C). The obtained solid residues were again centrifuged 2-3 times in 10 ml distilled water. Finally, purified pellets were dried at 60 °C in an oven to obtain powder and further used for FTIR analysis using potassium bromide (KBr) pellets. The pellets were measured at a range between 4000-400  $\text{cm}^{-1}$ . XRD analysis was conducted to determine the crystalline nature and phase purity of the AgNPs. The AFM samples were prepared by the hanging drop method, the AgNPs solution placed onto the glass slide to confirm morphology and distance of the nanoparticles. The HR-TEM analysis of AgNPs was carried out to confirm the size and surface morphology of AgNPs.

### Test treatment assay

Pea (*Pisum sativum* L.) seeds were obtained from the seed department of University of Agricultural Sciences (UAS), Dharwad. The seeds were surface sterilized 2-3 times with mercuric chloride to remove fungal contaminants and allowed to germinate on moistened filter paper in petridishes at room temperature. When the radical of the germinated seeds reached 0.5 to 1 cm, thirty seeds each were treated with different concentrations of AgNPs separately (4, 8, 12, 16  $\mu\text{g}/\text{ml}$ ) at the interval of 6, 12, 18 and 24 hours duration. The different concentrations of AgNPs solution were prepared from 100  $\mu\text{g}/\text{ml}$  stock solution. The experiments were conducted in triplicate. After completion of each treatment, roots from germinated seeds were immediately cut and fixed in Carnoy's fluid (ethyl alcohol: glacial acetic acid 3:1 v/v) for 24 hours. Then, they were transferred to 70% ethyl alcohol and stored in the refrigerator at 4 °C for further cytological studies.

### Scoring of cells

The cytological slides were prepared as suggested by Yanik et al. [49]. The stored root tips were hydrolyzed in 1 N HCl at 60 °C for 5-10 minutes and squashes were made with 2% aceto orcein for 10 min. Five slides were prepared for each treatment. A minimum of 600 cells were counted from both control and treated groups and approximately 1800 cells were scored for mitotic index and chromosomal abnormalities. The mitotic index and chromosomal abnormalities were calculated by using the following formula  $\text{MI} = \frac{\text{Number of dividing cells}}{\text{Total number of analyzing cells}} \times 100$  and chromosomal aberrations were calculated as  $\text{CAs} = \frac{\text{Number of abnormal cells}}{\text{Total no. of dividing cells}} \times 100$ . The cytological slides were photographed at 40X magnification with the bright field microscope (Carl Zeiss), assisted by (Axio imager M2) digital camera.

### Statistical analysis

The statistical analysis of data was done by using IBM statistical software (SPSS windows software version 20) followed by two-way ANOVA and Tukey test at the significance level of  $p < 0.05$ .

### Results and Discussion

#### UV-vis Spectroscopy

Silver nanoparticles were characterized by using UV-Vis spectroscopy. In the present investigation, the AgNPs were rapidly formed after the addition of 5 ml leaf extract to the 250 ml Erlenmeyer flask containing 95 ml of one millimolar silver nitrate ( $\text{AgNO}_3$ ) solution. The color of the reaction mixtures changes from yellowish to brown color which indicates the formation of AgNPs owing to the excitation of surface plasmon resonance [50,51]. The reaction mixture of AgNPs exhibits different absorbance peaks at 415 nm, 414 nm and 413 nm with pH 8, 9 and 10 respectively (Figures 1 and 2). Whereas, a blue shift represents the prominent peak of AgNPs at 413 nm with pH 10, which confirms the findings of previous studies [52-54]. It is evident from the results that the formation of AgNPs primarily depends on the pH of the reaction medium [55]. In basic and neutral pH the rate of formation of AgNPs is higher when compared to acidic pH.

#### Fourier Transform Infrared Spectroscopy (FTIR)

FTIR analysis reveals the involvement of functional groups present in the leaf extract of *B. stipularis* (Figure 3 and Table 1). The absorption peaks show IR bands at 3444.52, 2923.44, 2854.95, 1630.06, 1421.42, 1383.68, 1191.82, 1113.88, 1044.64 and 618.56  $\text{cm}^{-1}$ . The intense peak at 3444.52  $\text{cm}^{-1}$  corresponds to N-H/O-H stretching, the peak at 2923.94

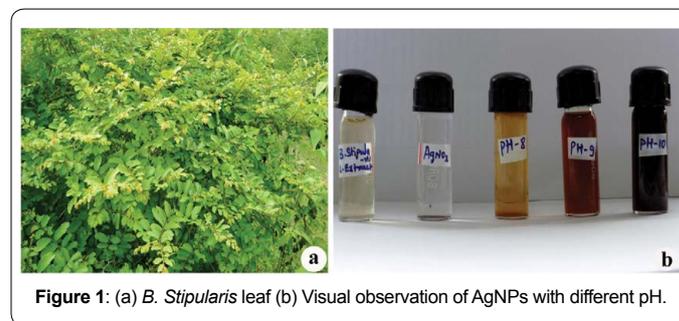


Figure 1: (a) *B. Stipularis* leaf (b) Visual observation of AgNPs with different pH.

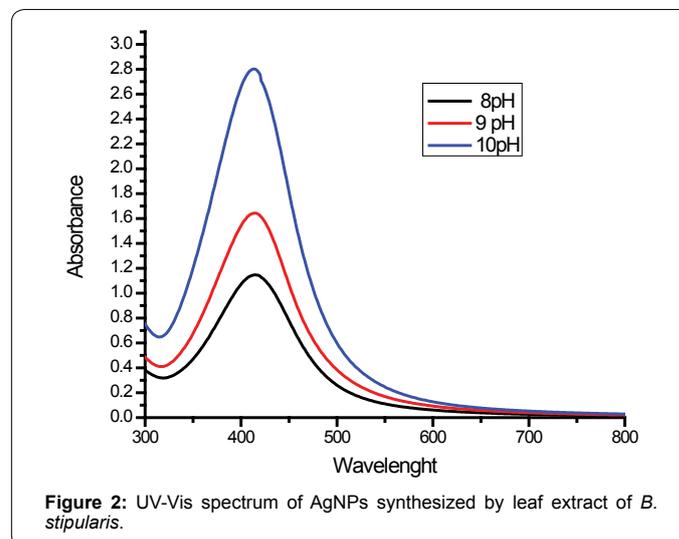


Figure 2: UV-Vis spectrum of AgNPs synthesized by leaf extract of *B. stipularis*.

cm<sup>-1</sup> corresponds to the methylene antisymmetric vibrational mode. The peak at 2854.95 cm<sup>-1</sup> is attributed to the presence of methylene symmetric vibrational mode. The band at 1630.06 cm<sup>-1</sup> represents amide band I and stretching of the carbonyl group. The bands at 1383.68 cm<sup>-1</sup> and 1191.82 cm<sup>-1</sup> corresponds -C-O- stretching of the carboxylation ions and -C-O-C- linkages respectively. The bands at 1044.64 cm<sup>-1</sup> and 618.56 cm<sup>-1</sup> corresponds C-S stretch (CH<sub>2</sub>-S) of thiol or thioether/-C-O-C- bonds and C=C group/ aromatic rings / C=O is stretching in carboxyl groups of proteins respectively. The present study confirms the presence of phytochemicals like alkaloids, flavonoids, tannins, proteins and carboxylic acids and was responsible for bioreduction, capping and stabilization of silver nanoparticles.

### X-ray Diffraction (XRD)

X-ray diffraction (XRD) studies were carried out to confirm the crystalline nature of AgNPs synthesized by leaf extract of *B. stipularis*. The diffraction peaks were observed at 2θ values of 38.2°, 64.6° and 77.5° which indexed to (111), (220) and (311) sets of lattice planes of face-centered cubic (fcc) of pure silver ions. The result confirms the Joint Committee on Powder Diffraction Standard (JCPDS) card number 04-0783. The FWHM value (111) plane was calculated to determine the mean size of silver nanoparticles using Debye-Scherrer's equation, which showed the crystalline nature of the silver nanoparticles and average size of 12.5 nm. Similar results were reported in AgNPs synthesized by using *Ginkgo biloba* leaf extract (Figure 4).

### Atomic Force Microscopy (AFM) and High Resolution Transmission Electron Microscopy (HRTEM)

The structure and surface morphology of silver nanoparticles were ascertained from the atomic force microscopy (AFM). The 2D and 3D images represent different size, shape and height distribution of the biosynthesized silver nanoparticles. The sizes of silver nanoparticles found to vary from 5 to 65 nm and are spherical in shape (Figure 5). The high resolution transmission electron microscopy (HR-TEM)

analysis of biogenic AgNPs shows the crystalline nature of the silver nanoparticles with a size range of 15 to 65 nm determined by the measurement scale bar (Figure 6).

### Effect of AgNPs on mitotic chromosomes of *Pisum sativum* L.

The cytotoxic and genotoxic effect of biogenic silver nanoparticles was evaluated on the basis of cytological parameters (mitotic index and chromosomal aberrations). Mitotic index is an important parameter and alternative method to determine growth inhibition and genotoxic effect of the cell cycle [56,57]. The root tip meristematic cells of *P. sativum* was exposed to different concentrations (4, 8, 12, 16 µg/ml) of AgNPs solution for 6, 12, 18 and 24 hours duration and results are presented in Table 2. The frequency of dividing cells was observed at all the stages of mitotic cell cycle such as prophase, metaphase, anaphase and telophase. The phase index of prophase is decreased frequently as compared to the control, which indicates the entering percentage of cells into the first phase of mitosis is decreasing [58]. Our data showed that the mitotic index of the control group was found to be highest at 55.43 ± 5.69% and lowest was in 13.99 ± 0.60% at 16 µg/ml concentration of AgNPs suspension for 24 hours duration of treatment, induced a significant reduction in a mitotic index. The results were found to be statistically significant at (p<0.05). Our investigation confirms the findings of a cyto-genotoxicity study of Yeken et al. [49] which strongly suggests that the exposure of *A. cepa* root tip cells to CPHE-AgNPs, CBE-AgNPs and AgS induce a significant decrease in mitotic index and increase various types of chromosomal aberrations in a dose and duration dependent manner (Figure 7). The decrease in the mitotic index might be due to the inhibition of cellular proliferative capacity of root cells, which causes slower progression of cells to DNA synthesis or S phase [59-61], blocking in the G<sub>2</sub>-phase of the cell cycle and averting the cell from entering mitosis [62] and completely destroying the metabolic activities of the cell, which leads to loss of viability and cause cell or DNA damage [63]. Akinboro et al. [64] reported that if the mitotic index is higher than the control, it can be harmful to the cells leading to disordered cell proliferation of meristematic cells. The interaction of silver nanoparticles with human

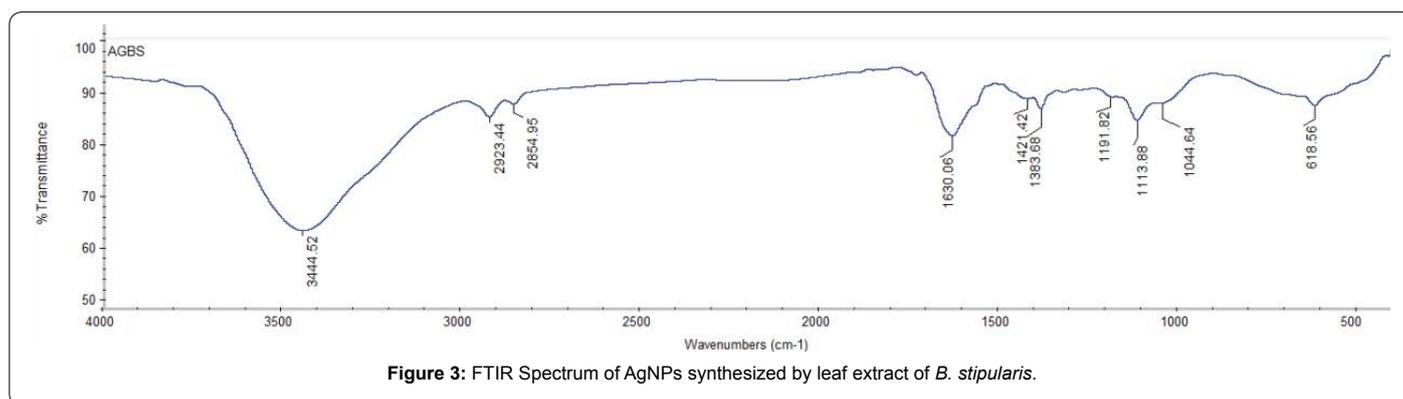


Figure 3: FTIR Spectrum of AgNPs synthesized by leaf extract of *B. stipularis*.

Sl. No.	Absorption peaks of AgNPs	Functional groups
1	3444.52	N-H/O-H stretching
2	2923.44	Methylene anti symmetric vibrational mode
3	2854.95	Methylene symmetric vibrational mode
4	1630.06	Amide band I/Carbonyl stretch
5	1383.68	-C-O- stretching of the carboxylation ions
6	1191.82	-C-O-C- linkages
7	1044.64	C-S stretch (CH <sub>2</sub> -S) of thiol or thioether/-C-O-C- bonds
8	618.56	C=C group/ aromatic rings/C=O is stretching in carboxyl groups of proteins

Table 1: FTIR absorption peaks and their associated functional groups.

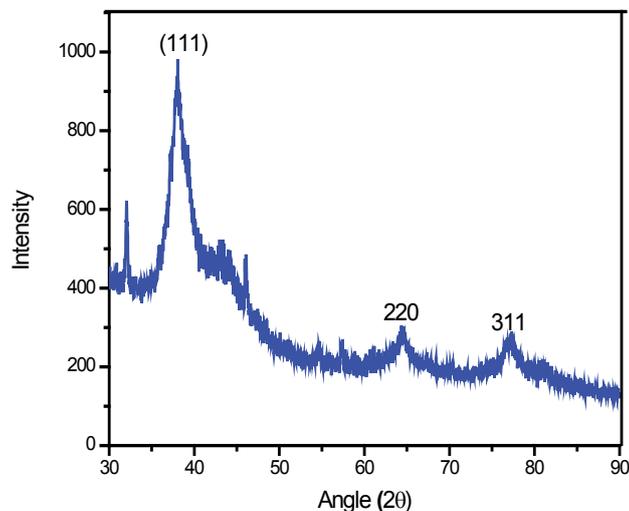


Figure 4: XRD Spectrum of AgNPs synthesized by leaf extract of *B. stipularis*.

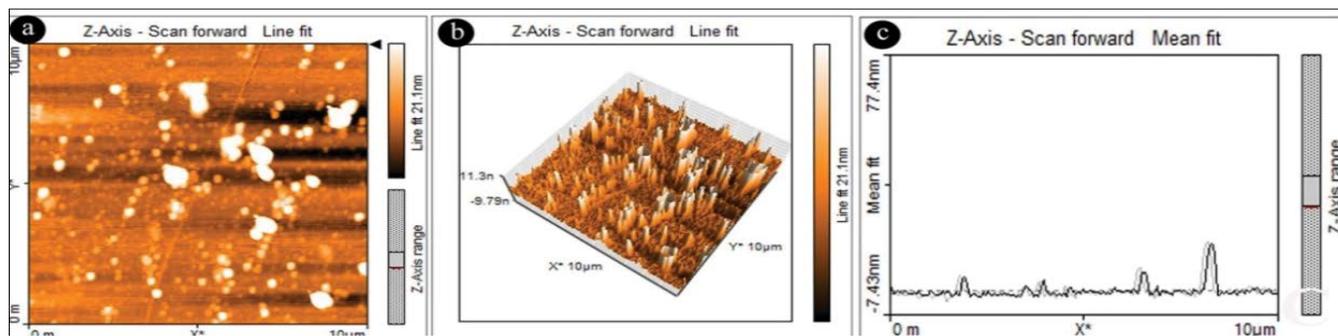


Figure 5: AFM image of AgNPs synthesized by leaf extract of *B. stipularis* (a) Two-dimensional image (b) Three-dimensional image (c) Topography image of AgNPs.

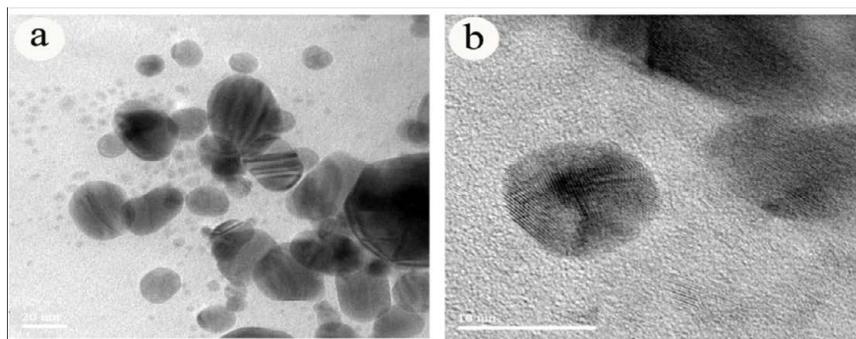


Figure 6: (a & b) HR-TEM image of AgNPs synthesized by leaf extract of *B. stipularis*.

serum albumin elicited binding affinity by bio-physical mechanism which causes significant modification of secondary structure, without any change in its basic structure [65]. Silver nanoparticles inhibited the growth of bacteria and initiate the lipid peroxidation reaction leading to glutathione depletion, disruption of membrane morphology and electron transport chain, which causes DNA damage [66,67]. The silver nanoparticles may bind to the bacterial cell membrane via hydrophobic and electrostatic interaction through pinocytosis, which promotes the degradation of cell wall [68]. The tiny sized nanoparticles have a greater surface area causing more toxicity when compared to larger sized nanoparticles [69]. When

these small sized nanoparticles enter into the bacterial cell membrane directly or indirectly interacting with nuclear material, resulting in DNA alteration and eventually inactivate the enzymes which are essential for ATP production and DNA replication [70-73]. Present findings confirm the reports of other researchers using different model plant systems viz. *Allium cepa*, [74-78], *Vicia faba* [79], *Triticum aestivum* [49], *Zea mays* [80] and *Drimia indica* [81].

The genotoxic effect of silver nanoparticles on apical meristems of root cells of *P. sativum* have been assayed on the basis of chromosomal

Durations (hrs)	Treatment of AgNP (µg/ml)	MI (%)	Chromosomal bridges	Chromosomal stickiness	Vacuolated nuclei	C- Metaphase	Multipolar	Micronucleus	CA (%)
	Control (DI)	55.43 ± 5.69	0.00 ± 0.00	0.00 ± 0.00	0.00 ± 0.00	0.00 ± 0.00	0.00 ± 0.00	0.00 ± 0.00	0.00 ± 0.00
6	4	48.16 ± 6.14	0.22 ± 0.01	1.37 ± 0.38	0.22 ± 0.03	0.11 ± 0.01	0.45 ± 0.21	0.44 ± 0.18	3.54 ± 0.18
	8	46.62 ± 4.44	0.36 ± 0.03	1.77 ± 0.92	0.36 ± 0.03	0.49 ± 0.05	0.77 ± 0.07	0.36 ± 0.03	5.03 ± 0.79
	12	43.75 ± 4.44	0.26 ± 0.02	2.55 ± 0.01	0.26 ± 0.02	0.82 ± 0.04	0.67 ± 0.06	0.28 ± 0.04	5.26 ± 0.97
12	16	42.06 ± 4.17	0.58 ± 0.02	1.59 ± 0.70	0.42 ± 0.04	0.57 ± 0.02	0.86 ± 0.04	0.45 ± 0.07	5.94 ± 1.71
	4	38.18 ± 2.76	0.30 ± 0.02	4.85 ± 1.76	0.29 ± 0.02	0.14 ± 0.02	0.14 ± 0.02	0.45 ± 0.04	7.25 ± 1.39
	8	36.04 ± 4.22	0.31 ± 0.02	4.40 ± 0.73	0.62 ± 0.02	0.31 ± 0.02	0.31 ± 0.02	0.46 ± 0.04	6.76 ± 0.68
	12	32.86 ± 1.87	0.66 ± 0.07	3.50 ± 0.56	0.50 ± 0.05	0.49 ± 0.01	0.82 ± 0.02	0.33 ± 0.02	7.00 ± 1.39
18	16	33.92 ± 4.46	0.70 ± 0.06	5.71 ± 0.48	0.51 ± 0.01	0.86 ± 0.02	1.55 ± 0.49	0.69 ± 0.03	10.39 ± 1.36
	4	31.95 ± 2.65	0.53 ± 0.05	8.04 ± 0.48	0.18 ± 0.03	0.71 ± 0.03	0.89 ± 0.03	0.70 ± 0.02	11.27 ± 0.30
	8	30.18 ± 2.47	0.38 ± 0.03	6.54 ± 1.67	0.18 ± 0.02	0.56 ± 0.05	1.33 ± 0.08	0.95 ± 0.03	11.12 ± 1.39
	12	26.32 ± 1.23	1.89 ± 0.05	6.37 ± 0.33	0.85 ± 0.05	0.83 ± 0.03	1.69 ± 0.39	0.83 ± 0.03	13.99 ± 0.47
24	16	24.28 ± 3.32	1.45 ± 0.07	8.57 ± 1.11	0.25 ± 0.04	1.48 ± 0.08	1.47 ± 0.74	0.24 ± 0.04	15.32 ± 1.34
	4	22.14 ± 0.85	0.78 ± 0.03	11.03 ± 1.43	0.27 ± 0.04	1.03 ± 0.04	0.78 ± 0.03	1.29 ± 0.41	16.78 ± 1.22
	8	19.82 ± 2.14	1.10 ± 0.43	10.35 ± 1.44	0.57 ± 0.09	0.84 ± 0.08	1.09 ± 0.95	1.65 ± 0.77	18.15 ± 1.02
	12	19.43 ± 0.80	1.91 ± 0.06	11.87 ± 3.16	1.59 ± 0.05	1.58 ± 0.51	3.82 ± 0.87	2.20 ± 1.37	25.90 ± 1.59
	16	13.99 ± 0.60	2.16 ± 0.02	16.54 ± 3.63	2.97 ± 0.79	2.19 ± 1.90	3.63 ± 0.46	4.32 ± 1.66	36.95 ± 3.15

Table 2: The effect of AgNPs on Mitotic Index (MI) and Chromosomal Aberrations (CAs) in *P. sativum* root tip meristematic cells.

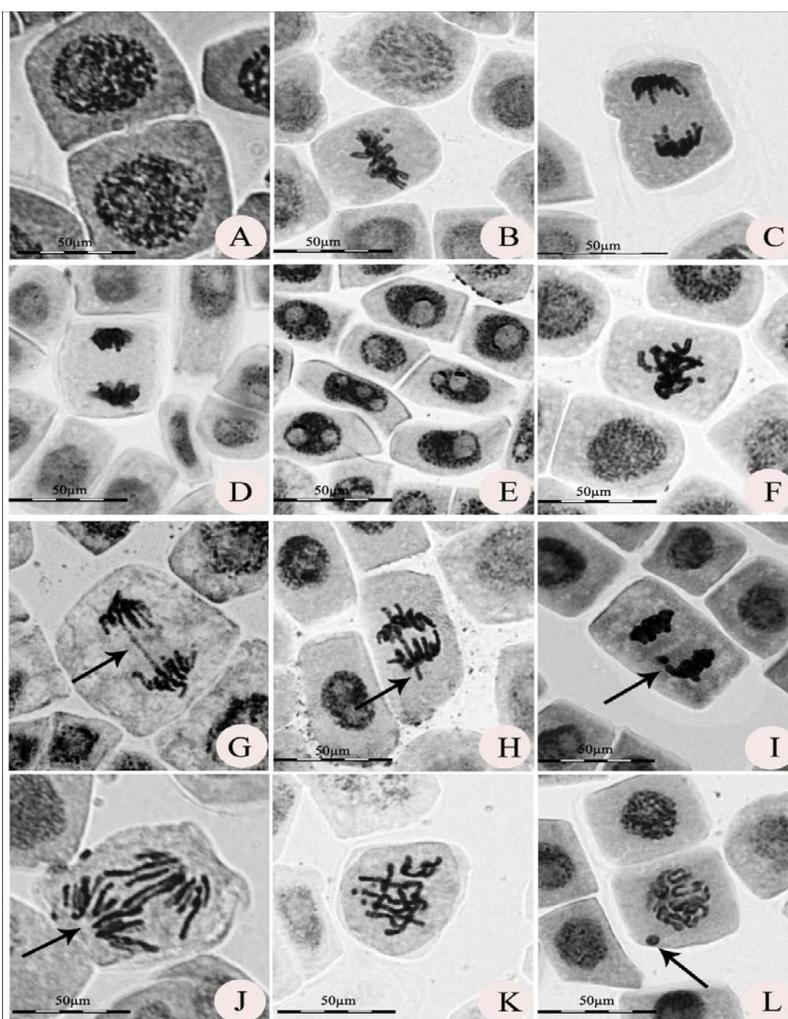


Figure 7: Effect of AgNPs on mitotic chromosomes of root meristem cells of *P. sativum* (A-D) Normal prophase, anaphase, metaphase and telophase (E) Vacuolated nuclei, (F) sticky metaphase, (G) single bridge at anaphase, (H) laggard, (I) telophase with fragment, (J) multipolar anaphase, (K) c-metaphase, (L) micronucleus.

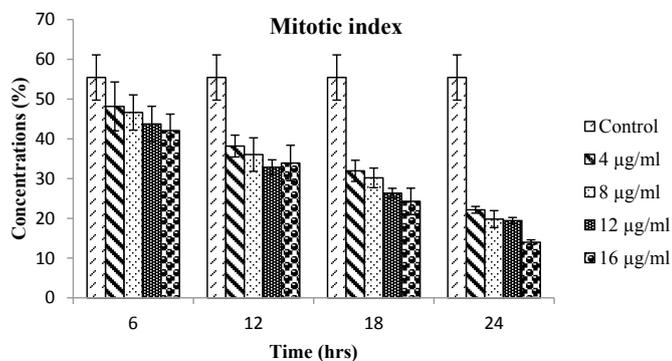
aberrations. There were no chromosomal aberrations in the control. An increase of chromosomal aberrations was recorded at 16  $\mu\text{g/ml}$  concentration ( $36.95 \pm 3.15\%$ ) of AgNPs for 24 hours duration of exposure. The results are statistically significant at ( $p < 0.05$ ) when compared to control. The induction of chromosomal aberrations was found to be higher at higher concentrations, which has affected cytoplasm and nucleus during cell division [82,83]. Studies of Manna et al. [64] showed an increase in toxicity of engineered nickel oxide nanoparticles on *Allium cepa* root cells along with an increase in genotoxicity at a very low dose.

The cyto-genotoxic effect of biogenic AgNPs was evaluated by analyzing various chromosomal aberrations viz. vacuolated nuclei at prophase, anaphase bridges, chromosomal stickiness, diagonal anaphase, c-metaphase, multipolar anaphase and micronucleus. Vacuolated nuclei were observed in both low and high doses of AgNPs suspension, it appeared that due to the deposition of nanoparticles in intercellular spaces of root cell resulting in chromatin loss and granulated nuclei become vacuolated [84,85]. The chromosomal bridges were occurring due to the presence of dicentric chromosomes and unequal exchange of sister chromatid segments undergoing translocation [86]. The chromosomal stickiness was the most frequent aberration in the present investigation, which arose due to the physiological stress in the form of plasmolysis. However, stickiness may be occurring due to malfunctioning of one or two types of specific nonhistone proteins, i.e. protein-protein interaction which makes chromatids to connect subchromatid bridges. El-Ghamery et al. [87], reported that the diagonal chromosomes may arise due to the effect of silver nanoparticles directly on spindle fibers resulting two sets of

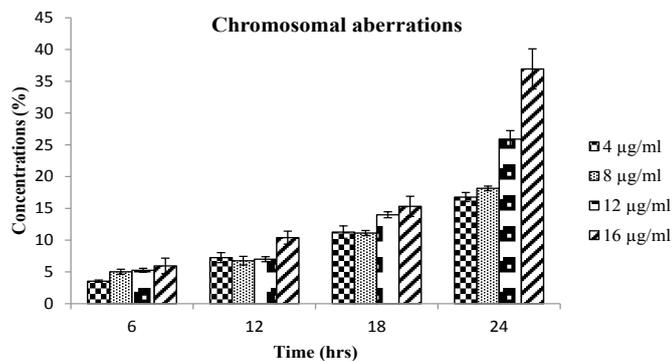
chromosomes which do not lie in the same position. The most common type of anomaly occurred in the cell cycle is c-metaphase induced by the interaction of nano-silver with mitotic spindles causing partial suppression of spindle formation and alteration of chromosomes at various stages of the cell cycle [88,89]. Micronucleus or small nuclei are formed from acentric chromosome fragments or whole chromosomes which are not integrated into the nucleus during cell division and are easily detectable in the ensuing stage of interphase [90,91]. It is evident from the present investigation that silver nanoparticles show inhibitory, mitodispersive and turbogenic effect on mitotic chromosomes of root tip meristematic cells of *P. sativum*.

## Conclusion

It has been concluded that the leaf extract of *B stipularis* is capable of producing silver nanoparticles from silver nitrate solution and these nanoparticles are relatively stable due to capping aptly by the alkaloids, flavonoids, tannins, proteins and carboxylic acids present in the leaf extract. The prepared biogenic silver nanoparticles are spherical, crystalline and monodispersed with an average size range between 5 and 85 nm. Thus the present study is a simple, safe, efficient, non-toxic and eco-friendly process. Furthermore, the green synthesized AgNPs could enter into the plant system directly causing DNA damage or cell damage in all the phases of cell cycle. The mitotic index was decreased ( $55.43 \pm 5.69\%$  to  $13.99 \pm 0.60\%$ ) at the 16  $\mu\text{g/ml}$  concentration for 24 hours duration of exposure. The present investigation revealed that the mitotic index is inversely proportional and chromosomal aberrations are directly proportional to the concentration and duration of exposure to silver nanoparticles (Figures 8 and 9).



**Figure 8:** The mitotic index (MI) in root tip meristematic cells of *P. sativum*, scored at different concentrations (4-16  $\mu\text{g/ml}$ ) and at different exposure periods (6-24 hours), found to be statistically significant at ( $p < 0.05$ ), when compared to the control.



**Figure 9:** The Chromosomal aberrations (CAs) in root tip meristematic cells of *P. sativum*, scored at different concentrations (4-16  $\mu\text{g/ml}$ ), and at different exposure periods (6-24 hours), found to be statistically significant at ( $p < 0.05$ ), when compared to the control.

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## Conflict of Interest

The author has none to declare.

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