

Review Article

Ammonium Alleviates Redox State in Solanum Seedlings under Cadmium Stress Conditions

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Abstract

Cadmium effects on growth and oxidative stress were investigated in 21-day-old tomato seedlings (*Solanum*) grown in hydroponics media containing 5 mM of KNO₃ or (NH4)₂SO₄ and three Cd levels as CdCl₂ (0, 5 and 25 μ M) for 14 d. Cadmium was more accumulated in nitrate-fed tomato compared to ammonium-fed ones. Dry weight, Chla, Chlb and carotenoides contents were reduced in NO₃-fed tomato. But in NH₄⁺-fed plants the parameters were increased. Cadmium induced an increase in the H₂O₂ and MDA levels which was more pronounced in nitrate-fed tomato. Antioxidant enzyme activities such as superoxide dismutase (SOD), catalase (CAT), and ascorbate peroxidase (APX) were induced with Cd. But, the enhancement degree of these enzymes activities were higher in ammonium-fed tomato compared to those grown with nitrate. These data suggested that antioxidative activity developed by tomato leaves is more induced by cadmium when ammonium was added in nutrient solution as nitrogen source. This can be related to the ability of cadmium to induce an accumulation of reactive oxygen species (ROS) less pronounced in presence of ammonium regime. The beneficial effect of NH₄⁺ on Cd toxicity was confirmed by a significant decrease in MDA level and accumulation of photosynthetic pigments.

Keywords: Antioxidant enzymes; Nitrogen; Tomato; Cadmium; Lipid peroxidation

Abbreviations: APX: Ascorbate Peroxidase; CAT: Catalase; H₂O₂: Hydrogen Peroxide; MDA: Malondialdehyde; NBT: Nitro Blue Tetrazolium; ROS: Reactive Oxygen Species; SOD: Superoxide Dismutase

Introduction

Cadmium (Cd) is one of the most important metals in terms of food-chain contamination, because it is readily taken up by the cells of different plant species [1,2]. In plants, Cd is known to disturb growth, amino-acid biosynthesis [3], nitrogen metabolism [4,5] and photosynthesis [6,7]. Plants face constant risk from reactive oxygen species (ROS), which are inevitably generated as products of photosynthesis and other cellular metabolic processes [8]. In plants, ROS are produced continuously as byproducts of various metabolic pathways that are localized in different cellular compartments [9], but under stressful conditions, their formation might be in excess of antioxidant scavenging capacity, thus creating oxidative stress by reaction and damage to all biomolecules, especially proteins, due to the higher rate constants of the reaction of the superoxide anion with amino acid side chains [10]. In addition, one of the most damaging effects of oxygen cytotoxic species and their products in cells is the peroxidation of membrane lipids [9]. High ROS levels can damage proteins and DNA [11]. Thus, plants need to earnestly control ROS overall levels by the co-ordinated action of several antioxidant enzymes such as superoxide dismutase (SOD, E.C. 1.15.1.1), catalase (CAT, E.C. 1.11.1.6), ascorbate peroxidase (APX, E.C. 1.11.1.11), and glutathione reductase (GR; EC 1.6.4.2). Superoxide dismutase is the major superoxide radical scavenger and its enzymatic action results in H₂O₂ and O₂ formation. The product of SOD activity (H₂O₂) is still toxic and must be eliminated by conversion to H₂O in subsequent reactions. CAT and several classes of peroxidases like APX then scavenge the H₂O₂ produced [9,12]. Inorganic nitrogen uptake into plant roots is under strict control in accordance with the nitrogen demand of the plant. Generally, plants prefers nitrate as nitrogen source. But, every other plant species grown in paddy fields predominantly utilizes ammonium during most of the growing period, since ammonium is the major form of inorganic nitrogen in hypoxic and anaerobic soils [13]. However, excessive ammonium uptake into plants can lead to toxic effects [5,14].

Among literature, oxidative stress is studied depending on the type of stress without return look at the form of nitrogen used. Since in nature, plants are in front of the various factors influencing his life. Study of antioxidative enzymes activity in plants cultivated in presence of different nitrogen forms and exposed to heavy metal might be for a great importance. In fact, today, environment pollution by heavy metals is a growing concern in the research community, since it may enter the environment through drainage water, river canal systems carrying industrial and different agricultural practices. Over that, ammonium nutrition is a widespread regime in the different ecosystems.

In the present research we essay to evidence that ammonium alleviates cadmium induced-oxidative stress in tomato leaves. In the same conditions, activities of antioxidant enzymes, which can be implicated in the oxidative stress defense, were measured.

Materials and Methods

Plant material

Seeds of the tomato (Solanum lycopersicon) were germinated in petri dishes in the dark. Seedlings were transferred and grown under

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continuous aeration in a nutrient solution containing 0.1 mM KNO₃. Plants were grown in a growth chamber under controlled conditions: a 16h-light (150 µmol m⁻² s⁻¹ PAR)/8h dark cycle, 22°C (light) /18°C (night) and 65% relatively humidity. The 7-day-old seedlings were supplied with the nutrient solution containing 5 mM of KNO₃⁻ or (NH₄)₂SO₄. After 14 days of metal exposure (0, 5 or 25 µM CdCl₂), leaves were harvested and used for chemical analyses.

Cadmium accumulation

Desiccated samples were ground to a fine powder using a porcelain mortar and pestle, then digested with an acid mixture $(HNO_3/HClO_4, 4/1 \text{ cm}^3/\text{cm}^3)$. Cd²⁺ concentration was determined by atomic absorption spectrophotometry (Perkin-Elmer, AAanalyst 300).

Photosynthetic Pigment estimation

Chla, Chlb and carotenoides contents were determined by the method of [15]. The absorbance of a sample was read at 645 and 663 nm. The pigment concentrations were calculated by equations allowing a simultaneous determination of Chla and Chlb, and carotenoids.

Soluble protein assays

Soluble proteins were measured after extraction of plant tissues (0.5 to 1g FW) at 4°C in 2 ml of H_2SO_4 (0.3 mM) and 0.5% (w/v) Polyclar AT. The homogenate was then clarified by centrifugation for 15 min at 30,000 g. Proteins were determined according to Bradford [16].

Estimation of lipid peroxidation

The malondialdehyde (MDA) content of leaves was determined by using the thiobarbituric acid method, as described by Alia et al. [17]. The leaves were homogenized in 5% (w/v) trichloroacetic acid (TCA). After centrifugation, a sample of the supernatant was added to 20% TCA containing 0.5% (w/v) thiobarbituric acid (TBA). The mixture was incubated at 95°C for 30 min. The concentration of thiobarbituric acid reacting substances was calculated using an extinction coefficient of 155 mM_1cm_1.

Measurement of H₂O₂

Content of $\rm H_2O_2$ in leaves, tissues was determined based on the modified method of [18]. $\rm H_2O_2$ contents were determined by colorimetric method from A508, using H2O2 (30% Sigma) (5-50 $\mu M)$ as a standard.

Antioxidative enzymes assays

Total CAT (EC 1.11.1.6) activity was assayed in presence of H_2O_2 , according to [19], by monitoring the decline in absorbance at 240 nm, as H_2O_2 was consumed. Enzyme activity was calculated using the extinction coefficient of 40 mM⁻¹ cm⁻¹ for H_2O_2 .

Total SOD (EC 1.15.1.1) activity was measured spectrophotometrically at 560 nm according to [20], based on the inhibition of the photochemical reduction of nitro blue tetrazolium (NBT). One unit of enzyme activity was defined as the quantity of SOD required for 50 % inhibition of NBT reduction.

Total APX (EC 1.11.1.11) activity was assayed in the presence of ascorbate by following the decline in absorbance of the oxidized ascorbate at 290 nm, according to [21]. Enzyme activity was calculated using the extinction coefficient of $2.8 \text{ mM}^{-1} \text{ cm}^{-1}$ for ascorbate.

Native polyacrylamide gel electrophoresis

Native-PAGE was performed in slab gels containing 4% acrylamide.

At the completion of electrophoresis, bands containing SOD activity were visualized with a tetrazolium assay [22]. After incubation at 30° C for 30 min in dark, the bands were revealed in phosphate tampon (50 mM) containing TEMED and riboflavine.

Statistical analysis

The results are the means \pm S.E. of at least three independent replicates. The analyses of variance were computed on statistically significant differences determined based on the appropriate F-tests. The mean differences were compared by utilizing Duncan's multiple range test.

Results

Dry weight production and Cd content

Cd effects on growth of tomato varied largely with nitrogen form. In fact, dry weight, leaf area and water tenors were negatively affected by Cd in NO_3 -fed plants. In contrary, in ammonium-fed tomato those parameters were not significantly affected by Cd (Figures 1A, 1B, 1C). Results shown in figure 1D demonstrated that cadmium was less accumulated in leaves of tomato grown with ammonium than with nitrate.

Soluble protein content

A significant effect on soluble protein content per DW was observed by both Cd and N regimes. A decrease in soluble protein content by Cd was observed when plants were grown with NO_3^- . However, plants grown under NH_4^+ regime and treated with Cd exhibited a significant increase of soluble protein content (Figure 1E).

Photosynthetic pigments contents

Photosynthetic pigments measurement showed that in control plants, Chla, Chlb and carotenoides were more accumulated in nitrate-fed tomato. Cd reduced pigments contents in nitrate grown plants and increased them in ammonium-fed tomato (Figure 2A, 2B, 2C).

Estimation of lipid peroxidation (MDA) and H₂O₂ production

Estimation of lipid peroxidation was determined in terms of the thiobarbituric (TBA)-reactive substances, such as MDA. In control plants, MDA content was higher in ammonium-fed tomato compared to those grown with nitrate. Regardless the nitrogen form, Cd exhibited an increase in MDA level (Figure 3A). But, the MDA level was more important in nitrate-fed plants.

 $\rm H_2O_2$ was much higher produced in tomato leaves received ammonium as nitrogen source (Figure 3B). Cd-treated tomato content increase $\rm H_2O_2$ tenor when Cd dose increased. Compared to ammonium-fed tomato, Cd induced a higher $\rm H_2O_2$ production in nitrate-fed plants (Figure 3B).

Antioxidant enzymes activities

In control plants, SOD activity was higher in tomato receiving ammonium than in those grown with nitrate. Regardless nitrogen form, Cd stimulated SOD activity in tomato leaves (Figure 4A). However, in ammonium-fed plants SOD activity was more important than in nitrate-fed ones.

In parallel, native gel electrophoresis of SOD showed that Cd induced four isozymes (a,b,c,d) in leaves of tomato plants (Figure 4B). But, the different isozymes were highly accumulated in ammonium-fed tomato compared to nitrate fed ones.

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(white symbol) as nitrogen source and treated with different doses of $CdCl_2$. Each point represents the mean \pm SD of triplicates from five independent experiments.

Catalase activity was more pronounced with ammonium than with nitrate (Figure 5A). Independently of nitrogen supplied, the activity of catalase was stimulated by Cd (Figure 5A). But in ammonium-fed tomato activity of this enzyme was higher than in nitrate grown plants.

A rise of the APX activity was observed in ammonium-fed tomato compared to those grown on nitrate (Figure 5B). In the tomato leaves, after cadmium exposure, APX activity increased in different leaves derived from different nitrogen regime.

Discussion

The environmental degradation, promoted mainly by anthropogenic action, has imposed strong pressure on the quality of ecosystems. The pollution of soil and water by a wide range of contaminants for both plants and animals has become a matter of great concern to researchers. In this sense, the elevated levels of heavy metals such as Cd in the environment are a reality today. Increasing Cd concentration in growth medium enhanced Cd accumulation in roots and shoot. However, tomato accumulated significantly higher Cd concentration in roots than in shoot [7]. Furthermore, depending on the nitrogen nutrient, potential differences among the plant so far analyzed have been observed in relation to their tolerance to Cd. In data shown here, there is evidence suggesting that tomato plants are partially protected against Cd when received ammonium as nitrogen source. In fact, when Cd was added in culture medium containing ammonium as nitrogen source, tomato plants showed better growth than those control, as dry weight and Leaves area remained unchanged. More that, water, Chla, Chlb and carotenoides contents were increased in those seedlings. The lower sensitivity of photosynthetic pigments in NH_4^{+} -fed tomato under Cd stress could be correlated to the lower content of Cd²⁺. This previous phenomena was explained by Chaignon

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et al. [23], suggesting that decline in absorption and accumulation of Cd^{2+} were probably the result of decline of pH of culture medium. It should be noted that in general, metal uptake is inhibited in acidic pH [24].

Despite being a non-redox metal, and thus not directly producing ROS [12], Cd can interfere with antioxidant defense systems. Under stressful conditions the protective system can be overridden by a rapid production of large amounts of ROS, leading to various structural modifications in proteins [25]. These oxidative modifications are characterized by the formation of carbonyl derivatives on side chains of histidine, arginine, and proline residues [26]. Our data demonstrated that the seedling received ammonium and exposure to Cd caused a remarkable increase in total soluble protein content in tomato seedlings. In another study, Cargnelutti et al. [25] showed that Hg-treated cucumbers presented increased total soluble protein content. To explain this result Verma and Dubey [27] suggested that the increase in protein content is possible due to de novo synthesis of stress proteins provoked by metal exposure. These stress proteins may constitute enzymes involved in GSH and phytochelatin biosynthesis and those required for Krebs cycle, as well as antioxidants and some heat shock proteins [28].



Figure 3: The level of lipid peroxidation (A) and H_2O_2 production (B), determined in terms of the TBA-reactive compounds (nmol MDA g-1 FW) in the leaves of tomato plants fed with nitrate (black symbol) or ammonium (white symbol) as nitrogen source and treated with different doses of CdCl₂. Each point represents the mean \pm SD of triplicates from five independent experiments.







triplicates from five independent experiments.

To estimate oxidative stress, the MDA tissue content had been widely used as an indicator of lipid peroxidation and, thereby, of oxidative damage in heavy-metal-exposed plants [29]. It was suggested that Cd may be involved in lipid peroxidation and membrane damage which was obvious from the significantly higher MDA content in leaves [30]. Our results in MDA contents showed that, regardless of nitrogen form used, presence of Cd induced a dramatic increase in the amount of lipid peroxidation product. But when fed with ammonium as nitrogen source, tomato accumulated lesser amount of MDA compared to in those grown with nitrate. This suggested that, NH⁺₄ reduced the untimely oxidative stress situation generally generated by Cd. The exposure of plants to Cd results in free radical production (H₂O₂, OH). This relationship between metallic stress and H₂O₂ production was well discussed in previous study [31]. To control the level of ROS and to protect the cells, plants possess antioxidant enzymes such as SOD, APX and CAT that scavenge ROS [9]. Superoxide dismutase, the first enzyme in the detoxifying process, converts superoxide radicals to H₂O₂ at a very fast rate [9]. The enhanced SOD activity observed in consistent with previous reports in which tomato and other plant species were treated with Cd [1,30]. Increase in SOD activity may be linked to an increase in superoxide radical formation as well as to de novo synthesis of enzyme protein [18,26], which in turn may be associated with an induction of genes of SOD by superoxide-mediated signal transduction [32]. In these data, we showed that the enhancement of SOD activity by Cd, shown in tomato derived from the two nitrogen regimes, was more pronounced in ammonium-fed tomato than in nitrate-fed seedlings. These results confirmed the fact that SOD plays an important role to alleviate oxidative stress by scavenging ROS from cell compartment. More that, it seems that NH4+-fed tomato were more stressed by Cd₂⁺, since SOD activity was dependent in the intensity of oxidative stress in plant cells [30]. However, this response suggested that antioxidant system might participate in protecting tomato biochemical structures against oxidative damages and minimizing the sensitivity of the photosynthetic machinery. In previous data we showed that photosynthesis process was more protected in Cd-stressed tomato when ammonium was used as nitrogen source [7].

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In another way, the increased CAT activity as found herein, which can be associated with H₂O₂ scavenging, was also observed by [1] in Coffea arabica and by [30] in Solanum lycopersicom under Cd-stress conditions. Independently of nitrogen form, this increase suggests a compensatory mechanism of defense against oxidative stress caused by this metal and can be explained by increase in its substrate to maintain the level of H2O2 as an adaptive mechanism of the plants [25]. Furthermore, the combined action of CAT and SOD is critical in mitigating the effects of oxidative stress, since their roles in the cell metabolism are complementary [12]. In this sense, it is interesting to note that both SOD and CAT activities increased in Cd treated tomato and that it is widely agreed that plants resist oxidative stress by increasing components of their intrinsic defensive system [12]. Another enzyme that can be activated to control and re-establish the homeostatic equilibrium of the redox status in cells is APX. In our study, the APX activity dynamics was similar to CAT and SOD in nitrate-fed tomato. Therefore, we suggest that also this enzyme participate to ROS scavenging. In fact, it was suggested that decline in $\mathrm{H_2O_2}$ level in pumpkin plants is mainly due to the scavenging action of APX [33]. In another hand, APX activity, a H₂O₂-scavenger that belongs to the ascorbate-glutathione cycle, was slightly decreased by Cd in ammonium regime case. It seems that the reduction in APX activity may be due to GSH depletion and a subsequent reduction in the ascorbate-glutathione cycle [1]. This reduction in GSH could be caused by an increased rate of phytochelatin synthesis induced by Cd ions as suggested by [1]. Ascorbate peroxidase could be responsible for the fine modulation of ROS for signaling, and its reduced activity would lead to a deleterious imbalance in ROS production and scavenging. Furthermore, the decreased activity of APX was apparently compensated for by the increased activity of other H₂O₂-degrading enzymes like CAT [34].

Our findings added new comparative information on the metabolism response of plants under heavy metal stress. These data indicated that regardless nitrogen form, the oxidative stress was induced with cadmium. But, it was more pronounced in ammoniumfed tomato compared to nitrate-fed ones. However, H2O2 and MDA contents were more important in nitrate-fed tomato upon addition of Cd.

We must emphasize that cadmium provoked a more important oxidative stress situation in tomato leaves cultivated with ammonium nitrogen and therefore suggest that antioxidative response was related to the intensity of oxidative stress induced rather than to the plant sensitivity to this metal. Also, it can be assumed that ammonium regime strongly protects Solanum Lycopersicum from Cd toxicity through reducing Cd uptake, and lipid peroxidation and improving the ROS scavenging antioxidant enzymes activities.

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