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## Removal of selected Pharmaceuticals from aqueous solutions using heterogeneous photocatalysis

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Pharmaceutical compounds (PhCs) in general characterized by their low biodegradability and high chemical stability, making conventional treatment technologies incapable to eliminate such kinds of recalcitrant compounds. In the current study, removal of two non-steroidal anti-inflammatory drugs (NSAID) from aqueous phase has been investigated through employing photolysis and heterogeneous photocatalysis processes.

Mefenamic acid (MEF) and diclofenac sodium (DCF) are commonly used as analgesic and inflammatory drugs and they are widely present in sewage systems. In this study, experements were divided into two parts: Photolysis and heterogeneous photocatalysis. Concerning photocatalysis experiments,  $TiO_2$  as a catalyst has been used in two forms: (i) as dispersed powder; (ii) immobilized on the surface of blue slabs. Obtained results showed that photolysis had low efficacy toward degradation of MEF (half-life (t1/2) 1442 min). While, using  $TiO_2$  as dispersed powder during the photocatalytic process enhanced the process dramatically and reduced half-life (t1/2) to 90 min, furthermore, it is possible to reach the complete mineralization after approximately 3 hours, whereas addition of  $TiO_2$  through immobilized system led to a little improvement in the photodegradation process behavior (t1/2 =1140 min.). For DCF photocatalysis by using  $TiO_2$  dispersed powder shows the fastest extent of degradation with 54.6 min of half-life (t1/2), while direct photolysis and photocatalysis using  $TiO_2$  immobilized system showed approximately comparable results (71 and 79 min respectively).

Kinetic studies for both drugs were accomplished and photoproducts were identified using liquid chromatography coupled with mass spectrometry system (LC-MS) (Fig.1).

The overall results suggested that using heterogeneous photocatalysis with  $TiO_2$  dispersed powder accelerate degradation process than  $TiO_2$  immobilized system. In spite of this, using  $TiO_2$  supported on glass substrates appears to be a promising alternative to conventional  $TiO_2$  suspension, since it is able to provide a clean method through saving a post treatment stage for recovering the catalyzer powder.