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JOINT EVENT

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Nano-scale membranes based on polystyrene sulfonic acid blends

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 $\mathbf{F}_{PSSA}^{lexible}$, strong, colorless and transparent polyvinylidenefluoride-olystyrenesulfonic acid (PVDF-PSSA) proton exchange membranes 2 are inverse in the indication of the strong strength of the strong strong strength of the strong strong strength of the strong s PSSA) proton exchange membranes1-3 were synthesized by first mixing DMF solutions of PVDF and tetrabutylammonium polystyrenesulfonate (PTBASS) terpolymer having smaller (<20%) mole fractions of styrene (S) and 4-chloromethylstyrene (CMS). The tetrabutylammonium styrene (BASS) sulfonate precursor monomer was synthesized through simple ion exchange of sodium styrene sulfonate and tetrabutylammonium-chloride or bromide. The PBASS terpolymer precursor was synthesized through a conventional radical copolymerization of BASS, styrene and CMS. The PVDF-PBASS blends are formed by dissolution of the two polymers in DMF, followed by evaporation and thermal cross-linking through annealing at about 170°C. The EAS type crosslinking appears to involve the formation of TBA benzyl sulfonates that may directly benzylate the S and CMS units of the terpolymer. The PVDF-PBASS blends are optically transparent above about 165°C and remain so after rapid cooling followed by ion exchange mediated by a mixture of 1.0 M H₃SO₂ and surfactants and repeated aqueous dialysis giving low (<40 wt%) water content PVDF-PSSA membranes. 4 Transmission electron microscopy confirms the presence of small (≥5 nm) PSSA domains the size and nature of which is still unknown. The water uptake, proton conductivity, ion exchange capacity (IEC), and methanol permeability of the membranes are tunable through variation of the PVDF/ PSSA content and the sulfonic acid content of the terpolymer. At higher PSSA contents (>20 wt.%) the membranes were shown to have proton conductivities comparable or higher than Nafion-117 but significantly lower methanol permeabilities. These low cost, environmentally appealing and conveniently accessible membranes can be used in direct methanol fuel cells (DMFCs) and other applications.



Recent Publication

1. Wang Y, Peng J, Li J (2017) PVDF based ion exchange membrane prepared by radiation grafting of ethyl styrene sulfonate and sequent hydrolysis. Radiation Physics and Chemistry 130:252-258.

Biography

Thieo E Hogen Esch has completed his PhD at the University of Leiden, Netherlands. Since then he has carried out research in the areas of physical organic chemistry and in the stereochemistry of anionic vinyl polymerization, the synthesis of block copolymers by anionic and free radical polymerizations at the University of Florida and since 1988 at the University of Southern California (USC). He has pioneered the synthesis and properties of "perfluorocarbon functionalized polymers" a topic that remains current. During the last few years his research has also included the synthesis and properties of thermodynamically unstable polymer blends of "camouflage mediated" homogeneous hydrophobic-hydrophilic type membrane.