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Synthesis and characterization of Cu/ZnO catalyst on CNTs

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Concentration of CO₂ in the atmosphere continues to rise due to various human activities. Catalytic conversion of CO₂ into valuable chemicals has attracted attention of many researchers around the world as it provides an alternative route to deal with CO₂ emission to atmosphere. In this study, Cu/ZnO catalyst is synthesized on multi-walled carbon nanotubes (CNTs) via incipient wetness impregnation method. The physicochemical properties of the catalysts have been analyzed using TEM, XRD, N₂ adsorption, H₂-TPR and XPS. The activity of the synthesized catalysts have been evaluated in a hydrogenation reaction of CO₂ using a fixed-bed micro reactor at 503 K and 22.5 bar. BET analysis revealed that the specific surface area of the sample decreased after deposition of metals onto the treated CNTs support. Observation via TEM showed agglomeration of nanoparticles on the exterior surface of the treated CNTs. Results of XPS analyses indicated the presence of Cu²⁺ in the catalyst sample. Evaluation of the synthesized catalyst in CO₂ hydrogenation reaction resulted in CO₂ conversion of 23% with selectivity of 73.7%, 2.0%, 21.9% and 2.4% for methyl formate, ethanol, methanol, and methane, respectively.

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Fullerene based organic and perovskite solar cells

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Among fullerene derivatives, PCBM offers the advantages of good solubility in organic solvents (chloroform, chlorobenzene, dichlorobenzene, etc.), higher electron mobility and higher electron affinity. However, weak absorption in the visible region and low lying LUMO level are the weak points. Weak absorption of PCBM limits the light harvesting in photovoltaic conversion and low LUMO level of the acceptor result in lower open circuit voltage (Voc) in PSCs, since Voc is strongly related to the difference between the LUMO level of acceptor and the HOMO of the donor material. Therefore, it is very important to design and synthesize new soluble fullerene derivatives with stronger visible absorption and higher LUMO energy levels than PCBM. It is crucial to control the lowest unoccupied molecular orbital (LUMO) of electron accepting materials for producing efficient charge transfer in bulk heterojunction (BHJ) solar cells. Due to their high LUMO level, soluble bis-adducts of C₆₀ are of high interest for improving the Voc in BHJ solar cells. In this work, we have developed novel fullerene derivatives for organic solar cells and perovskite solar cells. A novel bis-4-propylpentyl [6,6] methanofullerene bis-adduct, using an alkyl solubilizing group. The optoelectronic, electrochemical and photovoltaic properties of this bis-product are investigated. Perovskite heterojunction solar cells have attracted considerable attention because of their unique efficiencies. Novel fullerene bis-adduct dicarboxylic material showed good performance in perovskite solar cells. Novel Benzoic acid fullerene bis-adducts pay significantly more attention for engineering perovskite heterojunction solar cells to passivate the defects on surface and grain boundaries of perovskite films. Our photovoltaic results show that Benzoic acid fullerene bis-adduct compound is highly promising for the application in heterojunction perovskite solar cells because of its close solar cell efficiency to PCBM material. The carboxylic group may form hydrogen bond with I⁻ ion in the perovskite and passivate the surface of perovskite, thus reducing the recombination. Our results show that the efficiency of reference perovskite bulk heterojunction solar cell using PCBM is higher than 1.066 times that of perovskite heterojunction solar cells using benzoic acid fullerene bis-adducts. Our successful preliminary results suggested that further optimization of this novel fullerene bis-adduct can yield higher efficiencies with chemical modifications to fine tune the electronic properties.

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