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Kinetics of the cadmium uptake by a thiourea-modified natural clinoptilolite**Perfecto Barragan-Pena¹, Maria Teresa Olguin-Gutierrez² and Ma Guadalupe Macedo-Miranda³**¹Instituto Tecnológico de Nogales, Mexico²Instituto Nacional de Investigaciones Nucleares, Mexico³Instituto Tecnológico de Toluca, Mexico

To treat aqueous solutions with Cd²⁺ concentrations up to 67.83 mg/L experiments were carried out in a batch system where a natural modified zeolite from guaymas, Sonora, Mexico, was used as the adsorbent. The zeolitic material was modified with NaCl and thiourea. The experimentation was conducted at a pH=5.0. The results were compared to a study where the initial concentration for the zeolite modified with thiourea was 37.3 mg/L. Kinetics showed that the zeolite modified with NaCl and thiourea is best described according to pseudo-second order model, R²=0.9997, with a k₂=0.763 g/(mg h) and q_e=6.238 mg/g reached at 5 h. Compared to the non-modified adsorbent the thiourea-modified clinoptilolite improved its capacity to take up cadmium significantly.

Biography

Perfecto Barragan-Pena is currently a Professor at Instituto Tecnológico de Nogales, Mexico. He has a Doctorate in Environmental Sciences and has worked on sorption processes in both batch and dynamic systems, using natural adsorbents such as zeolites. Currently he teaches sustainable development and inferential statistics.

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